

SCHOOL INTERNSHIP

LESSON PLAN DIARY

CUM

JOURNAL

B.Ed. Session 2018-2020

Name of Intern	JULIE KOSHY
Enrolment No.	18021010278
Subject	SCIENCE

SCHOOL OF EDUCATION
(SOE)



(Established under Galgotias University Uttar Pradesh Act No. 14 of 2011)

Plot no.: 2, Sector: 17A, Yamuna Expressway, Gautam Budh Nagar, UP (India) 203201

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INDEX

Lesson Plan No.	Class & Sec.	Topic	Date
1	IX-A	MATTER & ITS CLASSIFICATION	13.01.2020
2	VII-C	ACIDS & BASES	14.01.2020
3	VII-B	HEAT	15.01.2020
4	VII-B	TRANSFER OF HEAT	16.01.2020
5	VIII-D	SYNTHETIC FIBRES	16.01.2020
6	VII-A	PHYSICAL & CHEMICAL CHANGES	20.01.2020
7	VI-B	LIGHT, SHADOWS & REFLECTION	20.01.2020
8	VI-B	LIGHT, SHADOWS & REFLECTION	21.01.2020
9	VI-A	SEPARATION OF SUBSTANCES	22.01.2020
10	VI-A	SEPARATION OF SUBSTANCES(SOLID FROM OTHER SOLIDS)	23.01.2020
11	VI-A	SEPARATING INSOLUBLE SOLIDS FROM LIQUIDS	24.01.2020
12	VI-C	THE LIVING ORGANISMS & THEIR SURROUNDINGS	25.01.2020
13	VI-D	MOTION & ITS TYPES	27.01.2020
14	VI-C	ADAPTATION IN ANIMALS AND PLANTS	28.01.2020
15	VI-C	ADAPTATION IN ANIMALS AND PLANTS	29.01.2020
16	X-D	CHEMICAL EQUATIONS	30.01.2020
17	X-A	TYPES OF CHEMICAL REACTIONS	31.01.2020
18	X-C	OXIDATION & REDUCTION REACTIONS	3.02.2020
19	VI-C	ADAPTATION TO GRASSLANDS	4.02.2020
20	VI-C	AQUATIC ADAPTATION IN ANIMALS & PLANTS	5.02.2020
21	VI-B	ELECTRICITY	6.02.2020
22	VI-B	LIGHTING UP A BULB WITH A DRY CELL	7.02.2020
23	VI-B	ELECTRIC CIRCUITS	10.02.2020
24	VIII-D	PLASTICS	11.02.2020
25	VII-A	REFLECTION OF LIGHT	12.02.2020

Galgotias University Uttar Pradesh, G.B. Nagar
School of Education

Lesson Plan No. 1

Date 13/01/20

Name of the Internship School National Victor Public School

Class & Section IX-A Period 2nd Duration 40 mins

Subject Science Sub-Subject Chemistry

Topic Matter and its classification

Specific Objectives After my class, students will be able to

- define matter
- state the characteristics of particles of matter.
- classify matter into 3 states.

Previous Knowledge

Students have general idea about atoms, molecules and intermolecular forces of interaction

Introduction

Teacher Activity	Student Activity
1. What is this? (showing book)	Book
2. If I place this book on table, what does it occupy?	Space
3. If I fill this bottle with water, what does it occupy?	Space
4. So, anything that occupies space is called _____.	No Response



Announcement of the Topic

Today we will discuss about 'Matter and its classification'.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Meaning and definition of Matter.	Student defines matter	<p><u>Teacher's Statement</u></p> <p>Anything that occupies space and has mass is called matter. The air we breathe, the food we eat, stars, plants, animal and even a small drop of water or a particle of sand - everything is matter.</p>	Student notes down the definition of matter in their notebook.	Lecture Method.	
2. Characteristics of Particle nature of Matter	Student states the characteristics of Particle nature of matter.	<p><u>Teacher's Statement</u></p> <ol style="list-style-type: none"> Matter is made up of tiny particles. Vacant spaces are present in particles of matter. Particles of matter are in continuous motion. Particles of matter are held together by forces of attraction. 	Student listens carefully and notes down in their notebooks.	Inductive Method Lecture Method	Define Matter.
3. Classification of Matter	Student classifies matter based on particle arrangement.	<p><u>Ques:</u> What are the three states of matter?</p> <p><u>Statement:</u> Based on particle arrangement matter is classified into</p> <ul style="list-style-type: none"> - Solid state - Liquid state - Gas state. 	Student listens carefully and writes down in their notebooks.	Lecture Method. <u>Boardwork</u> Class: IX Subject: Science Topic: MATTER AND ITS CLASSIFICATION Based on particle arrangement matter is classified into - Solid state - Liquid state - Gas state	State the characteristics of particle nature of matter. Classify matter into 3 states.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
4. Properties of Solid State	Student lists the properties of Solid State.	<p><u>Teacher's Statement</u></p> <p>Solid state has the following properties :-</p> <ol style="list-style-type: none"> Rigid Fixed shape and definite volume. Strong intermolecular forces of attraction between the particles. <p>eg: Dust, chalk, pen etc.</p>	Student listens carefully and notes down in their notebook.	<p>Boardwork</p> <p>Class: IX Subject: Science</p> <p>Topic: MATTER AND ITS CLASSIFICATION</p> <p><u>SOLID STATE</u></p>  <p>→ Rigid → Fixed Shape & Volume → Strong Intermolecular forces of attraction</p>	list the properties of Solid state.
5. Properties of Liquid State	Student lists the properties of liquid state	<p><u>Ques:</u> Give some examples of liquids.</p> <p><u>Statement</u></p> <p>Liquids include water, milk, wine, blood etc.</p> <p>Liquid state has the following properties :-</p> <ol style="list-style-type: none"> No fixed shape. Has fixed volume. Less intermolecular force of attraction. 	Water, Juice Student notes down the characteristics of liquid state in their notebook.	<p>Lecture Method</p> <p>Class: IX Subject: Science</p> <p>Topic: MATTER AND ITS CLASSIFICATION</p> <p><u>LIQUID STATE</u></p>  <p>→ No fixed shape → Has fixed volume → Less intermolecular force of attraction</p>	list the properties of liquid state.
6. Properties of Gaseous state	Student states the properties of gaseous state.	<p><u>Teacher's Statement</u></p> <p>Gaseous state has the following properties :-</p> <ol style="list-style-type: none"> No fixed shape. No fixed volume. Very less / no intermolecular force of attraction. <p>eg: Air, CO₂, N₂, O₂, O₃ etc.</p>	Students listen carefully and write down in their notebook.	<p>Lecture Method Deductive Method</p>	State the properties of gaseous state.

Closure

Ques: What are the characteristics of the particle nature of matter?

Ques: What are the different states in which matter can be classified?

Ques: State 3 differences between solid, liquid & gaseous state.

Ques: Give 3 examples of solid, liquid and gas state.

Home Assignment

Give Reasons:-

The smell of hot sizzling food reaches you several metres away, but to get the smell from cold food, you have to go close.

Signature of the Mentor

Assessment of Lesson

Name of Intern: Julie Koshy Date 13/01/2020
 Topic: Matter and its Classification Lesson Plan No. 1

Category Under Supervision		Rating Scale					
		Not Used	Poor	Avg.	Good	Very Good	
A.	Lesson Plan	1. Objectives stated in behavioural terms	0	1	2	3	4
B.	Introduction	2. Student's previous knowledge tested	0	1	2	3	4
		3. Topic was announced effectively	0	1	2	3	4
C.	Presentation	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
		5. Skill of Questioning	0	1	2	3	4
		6. Skill of Explanation	0	1	2	3	4
		7. Skill of Stimulus variation	0	1	2	3	4
		8. Skill of Reinforcement	0	1	2	3	4
		9. Skill of Illustration with examples	0	1	2	3	4
D.	Use of Teaching Aid	10. Adequate & appropriate	0	1	2	3	4
		11. Quality of teaching aid	0	1	2	3	4
E.	Chalk Board Work	12. Organized and adequate	0	1	2	3	4
		13. Appropriateness and clarity	0	1	2	3	4
F.	Management of the Class	14. Class environment & discipline	0	1	2	3	4
		15. Supervision of student's work	0	1	2	3	4
G.	Evaluation	16. Skill of Closure	0	1	2	3	4
		17. Based on the behavioral objectives	0	1	2	3	4
		18. Quality of home assignment	0	1	2	3	4

Over all comments and suggestions:-
 (Dress Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of Lesson with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Journal

(Description of self experience by the Intern after teaching)

Category Under Supervision			Rating Scale			
			Not Used	Poor	Avg.	Good
A.	Introduction	1. Student's previous knowledge tested	0	1	2	3
		2. Topic was announced effectively	0	1	2	3
B.	Presentation	3. Systematic & logical development of the concept with the help of pupil	0	1	2	3
		4. Skill of Questioning	0	1	2	3
		5. Skill of Explanation	0	1	2	3
		6. Skill of Stimulus variation	0	1	2	3
		7. Skill of Reinforcement	0	1	2	3
		8. Skill of Illustration with examples	0	1	2	3
		9. Proper use of teaching aid(s)	0	1	2	3
C.	Use of Teaching Aid	10. Tried to create interest in learning	0	1	2	3
		11. Organized and adequate	0	1	2	3
D.	Chalk Board Work	12. Appropriateness and clarity	0	1	2	3
		13. Class environment & discipline	0	1	2	3
E.	Management of the Class	14. Supervision of student's work	0	1	2	3
		15. Skill of Closure	0	1	2	3
F.	Evaluation	16. Based on the behavioral objectives	0	1	2	3
		17. Quality of home assignment	0	1	2	3

Over all experience and suggestion for my future teaching from myself

(Eg.- I satisfied the queries of the students. Today I feel satisfied/unsatisfied with my teaching.)

Julie

Signature of the Intern

Signature of the Mentor

Galgotias University Uttar Pradesh, G.B. Nagar
School of Education

Lesson Plan No. 2

Date 14.01.2020

Name of the Internship School National Victor Public School

Class & Section VII-C Period 2nd Duration 40 mins

Subject Science Sub-Subject Chemistry

Topic Acids and Bases

Specific Objectives After my class, students will be able to

- differentiate between an acid and a base
- classify acid on basis of their source
- define indicators.

Previous Knowledge

Students have general idea about different eatables and their tastes.

Introduction

Ques: What comes into your mind when you see these?
(showing lemons, lemon, orange, bitterguard)

Ans: They are eatables and have different tastes.

Ques: What is the taste of lemon, curd, vinegar?

Ans: They are sour in taste.

Ques: What about bitterguard?

Ans: Bitter.

Ques: Why are their tastes different?

Ans: No Response


Announcement of the Topic

So, students, today we'll discuss about "Acids & Bases".

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Acid	Student classifies acid on the basis of source from which it is available.	<p><u>Teacher's Statement</u></p> <p>The word acid comes from the latin word 'acere' which means sour. So, the substances that are sour in taste are called acids.</p> <p>eg: curd contains lactic acid, lemon juice contains citric acid, vinegar contains acetic acid.</p> <p><u>Classification of Acids</u></p> <p>Acids are classified into 2 -</p> <p>i) Organic Acid: Acids that come from plants, animals or human.</p> <p>eg: Lactic Acid - Curd Citric Acid - Oranges Oxalic Acid - Tomatoes Ascorbic Acid - Apples Formic Acid - Bee sting</p> <p>ii) Mineral Acid: Acids that come from minerals.</p> <p>eg: HCl, H₂SO₄, HNO₃</p>	Student listens carefully and notes down in their notebook.	<p>Lecture Method</p> <p>Boardwork</p> <p>Class: VII Subject: Science Topic: ACIDS AND BASES</p> <pre> ACID / \ ORGANIC MINERAL ACID ACID - LACTIC ACID - HCl - CITRIC ACID - HNO₃ - OXALIC ACID - H₂SO₄ - ASCORBIC ACID - FORMIC ACID </pre>	Classify acid on the basis of source from which it is available.
2. Base	Student classifies base as strong and weak.	<p><u>Teacher's Statement</u></p> <p>Substances which are bitter in taste and feel slippery on touching are known as bases and the nature of such substances is said to be basic.</p> <p>Strong Base :- Corrosive, can't be consumed. eg: Sodium hydroxide, potassium hydroxide.</p> <p>Weak Base :- Very mild & can be consumed. eg: Ammonium hydroxide (Toothpaste), Magnesium hydroxide (Milk of magnesia).</p>	Student listens carefully and notes down in their notebook.	Lecture Method	Classify base into 2.

Presentation

Teaching Point	Behavioural Objectives	Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
3. Indicators	Student states the use of indicators.	<u>Teacher's Statement</u> Substances that are used to determine whether a substance is acidic or basic are known as Indicators. These indicators change their colour when added to a solution containing an acidic or a basic substance. eg: Turmeric, litmus, china rose petals - some naturally occurring indicators.	Student listens carefully and notes down the different types of indicators in their notebook.	Lecture Method Demonstration Method.	
o Meaning					
o Types				Class: VII Subject: Science Topic: ACIDS AND BASES	
- Litmus paper				LITMUS TEST :- <div style="display: flex; align-items: center;"> <div style="border: 1px solid black; padding: 2px; margin-right: 5px;">R</div> <div style="margin-right: 5px;">TURNS RED WHEN TESTED ON ACID</div> </div> <div style="display: flex; align-items: center;"> <div style="border: 1px solid black; padding: 2px; margin-right: 5px;">B</div> <div style="margin-right: 5px;">TURNS BLUE WHEN TESTED ON BASE</div> </div> <p>CHINA ROSE INDICATOR</p> <div style="display: flex; align-items: center;">  <div style="margin-left: 5px;">ACIDIC SOLUTION TURNS PINK</div> </div>	
- China Rose		Litmus Test :- Litmus is the most commonly used natural indicator available as strips of paper or solution. It is extracted from lichens. It turns red on addition of an acidic solution. It turns blue on addition of a basic solution.			
- Phenolphthalein		China Rose Indicator :- Another natural indicator which turns acidic solutions to dark pink or magenta colour and basic solution to green colour. Phenolphthalein :- It is another indicator giving pink colour in case of basic solution and remains colourless in case of acidic solution.			State the use of different types of indicators.

Closure

So students, today we discussed about

- Acid : Mineral Acids & Organic Acids

- Base : Strong & Weak Base

- Indicators : litmus, China Rose, Phenolphthalein

Ques : Differentiate between mineral acid & organic acid.

Ques : Name the source from which litmus solution is obtained. What is the use of this solution?

Home Assignment

Ques : Explain why :-

- Antacid tablet is taken when one suffers from acidity.
- Calamine solution is applied on the skin when an ant bites.

Assessment of Lesson

Name of Intern: Julie Koshy Date: 14.01.2020
 Topic: Acids and Bases Lesson Plan No. 2

Category Under Supervision		Rating Scale				
		Not Used	Poor	Avg.	Good	Very Good
Lesson Plan	1. Objectives stated in behavioural terms	0	1	2	3	4
	2. Student's previous knowledge tested	0	1	2	3	4
	3. Topic was announced effectively	0	1	2	3	4
Introduction	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
	9. Skill of Illustration with examples	0	1	2	3	4
Presentation	10. Adequate & appropriate	0	1	2	3	4
	11. Quality of teaching aid	0	1	2	3	4
Use of Teaching Aid	12. Organized and adequate	0	1	2	3	4
	13. Appropriateness and clarity	0	1	2	3	4
Chalk Board Work	14. Class environment & discipline	0	1	2	3	4
	15. Supervision of student's work	0	1	2	3	4
Management of the Class	16. Skill of Closure	0	1	2	3	4
	17. Based on the behavioral objectives	0	1	2	3	4
Evaluation	18. Quality of home assignment	0	1	2	3	4

All comments and suggestions:-

(Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of work with the help of Students)

Julie
 Signature of the Intern

Signature of the Mentor

Signature of the

Galgotias University Uttar Pradesh, G.B. Nagar
School of Education

Lesson Plan No. 3

Date 15/01/2020

Name of the Internship School National Victor Public School

Class & Section VII-B Period 2nd Duration 40 mins

Subject Science Sub-Subject Physics

Topic Heat

Specific Objectives After my class, students will be able to

- distinguish between heat & temperature
- cite the different types of thermometers.
- explain the correct method of reading a clinical thermometer.

Previous Knowledge

Students have general idea about hot and cold objects.

Introduction

Ques: How do we know whether an object is hot or cold?

Ans: By touching the object.

Ques: But, is our sense of touch reliable?

Ans: Not always.

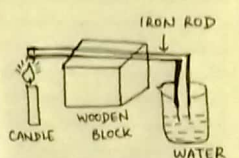
Ques: So apart from touching, how do we measure the hotness/coldness of an object?

Ans: No Response.

Announcement of the Topic

So students, today we'll discuss about "Heat".

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Heat and temperature	Student distinguishes between heat and temperature.	<p><u>Teacher's Statement</u></p> <p>Heat is a form of energy that is transferred from a system to its surroundings or from one object to another as a result of a difference in temperature. Heat is measured in Joules (J).</p> <p>Temperature is a measure of how hot or cold a substance is. It is measured in degree Celsius (°C).</p> <p>We use a thermometer to measure the temperature of an object or substance.</p> <p>Pupil teacher performs the following activity :-</p> <ul style="list-style-type: none"> o A beaker filled with some fresh water is placed. o Asks the students to touch the beaker. o Places an iron rod on a wooden block and inserts its one end in the beaker. o Lights a candle under the other end of the rod. o After some time, asks the students to touch the water again. o How has it become warmer? <p>Therefore, heat is a form of energy transferred from hot body to cold body. Heat changes the temperature of an object.</p>	<p>Student listens carefully and notes the difference between heat and temperature in their notebook.</p> <p>Student follows the instructions:</p> <p>The water is cold in the beaker.</p> <p>The water is warmer than it was before.</p> <p>Heat energy is transferred from hot end to the cold end of the iron rod, so water in the beaker becomes warmer.</p>	<p>lecture Method</p> <p>Demonstration Method</p> <div style="border: 1px solid black; padding: 5px; margin: 10px 0;"> <p>Class: VII Subject: Science Topic: HEAT</p>  <p>HEAT IS TRANSFERRED FROM HOT END TO COLD END OF THE IRON ROD, SO WATER BECOMES WARMER.</p> </div>	<p>Distinguish between heat and temperature.</p>

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
2. Types of thermometers	Student cites the different types of thermometers	<p><u>Teacher's Statement</u></p> <p><u>Clinical thermometer</u> It is used to measure human body temperature in the range of 35°C to 42°C. It consists of a long, narrow, uniform glass tube. It has a bulb at one end which contains mercury. Outside the bulb a small shining thread of mercury can be seen.</p> <p><u>Laboratory thermometer</u> It is used to measure the temperature of hot solids & liquids in experiments. It measures temperature in the range of 10°C to 110°C and on high temperatures.</p> <p><u>Digital thermometer</u> These are advanced thermometers used to measure the temperature of a body with a high level of accuracy.</p>	Student listens carefully and notes down the difference between the three types of thermometers	<p>Lecture Method</p> <p>Class: VII Subject: Science Topic: HEAT</p> <p><u>TYPES OF THERMOMETERS</u></p> <ul style="list-style-type: none"> CLINICAL THERMOMETER (35°C - 42°C) (Temp of human body) LABORATORY THERMOMETER (-10°C - 110°C) (Temp of hot solids/liquids) DIGITAL THERMOMETER 	
3. Reading a thermometer (clinical thermometer)	Student explains the correct method of reading a clinical thermometer	<p><u>Teacher's Statement</u></p> <ul style="list-style-type: none"> → Wash the thermometer properly with an antiseptic solution → Hold it definitely and give a few jerks which will bring the level of mercury down below 35°C. → Place the bulb of the thermometer under the tongue. → Wait for a minute. Take the thermometer out & note the reading. → This reading gives the exact body temperature. 	Student notes down in their notebook	Lecture Method	<p>Cite the different types of thermometers</p> <p>Explain the correct method of reading a clinical thermometer.</p>

Closure

So students, today we discussed about

- heat & temperature
- different types of thermometers
- correct method to read a clinical thermometer

Ques : State the similarities & differences between laboratory and clinical thermometer.

Ques : Fill in the blanks :-

- i) The hotness of an object is determined by _____.
- ii) Temperature is measured in degree _____.
- iii) Temperature of boiling water cannot be measured by a _____ thermometer.

Home Assignment

Ques : An iron ball at 40°C is dropped in a mug containing water at 40°C . The heat will

- a) flow from iron ball to water or
- b) flow from water to iron ball. or
- c) increase the temperature of both. Give reasons.

Assessment of Lesson

Name of Intern: Julie Koshy Date 15/01/2020
 Topic: Heat Lesson Plan No. 3

Category Under Supervision		Rating Scale				
		Not Used	Poor	Avg.	Good	Very Good
Lesson Plan	1. Objectives stated in behavioural terms	0	1	2	3	4
Introduction	2. Student's previous knowledge tested	0	1	2	3	4
	3. Topic was announced effectively	0	1	2	3	4
Presentation	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
	9. Skill of Illustration with examples	0	1	2	3	4
Use of Teaching Aid	10. Adequate & appropriate	0	1	2	3	4
Chalk Board Work	11. Quality of teaching aid	0	1	2	3	4
	12. Organized and adequate	0	1	2	3	4
Management of the Class	13. Appropriateness and clarity	0	1	2	3	4
	14. Class environment & discipline	0	1	2	3	4
Evaluation	15. Supervision of student's work	0	1	2	3	4
	16. Skill of Closure	0	1	2	3	4
	17. Based on the behavioral objectives	0	1	2	3	4
	18. Quality of home assignment	0	1	2	3	4

For all comments and suggestions:-

Lesson Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of lesson with the help of Students)

Julie
 Signature of the Intern

Signature of the Mentor

Signature of the Mentor

Galgotias University Uttar Pradesh, G.B. Nagar
School of Education

Lesson Plan No. 4

Date 16/01/20

Name of the Internship School National Victor Public School

Class & Section VII-B Period 4th Duration 40 mins

Subject Science Sub-Subject Physics

Topic Transfer of Heat

Specific Objectives After my class, students will be able to:

- state the 3 ways in which heat can be transferred from one object to another.
- describe the process of conduction.
- cite the difference between convection and radiation.

Previous Knowledge

Students have good knowledge about heat and temperature.

Introduction

Ques: What happens when a frying pan is kept on a flame?

Ans: It becomes hot.

Ques: Why so?

Ans: It is because heat passes from the flame to the utensil.

Ques: What happens when this pan is removed from fire?

Ans: It cools down.

Ques: What is this process of heat transfer referred to as?

Ans: No Response.

Announcement of the Topic

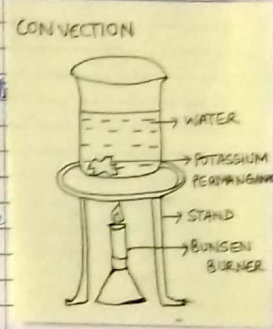
"Today, we'll discuss about "Transfer of Heat".

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
Transfer of Heat	Student cites the 3 ways of transfer of heat.	<p>Pupil teacher does the following activity :-</p> <ul style="list-style-type: none"> An aluminium rod is taken Four small wax pieces are stuck on the rod at equal distance The rod is clamped to a stand Now, one end of the rod is heated. <p>Q: What happens to the wax pieces? Q: Which piece falls first?</p> <p>So, heat is transferred from the end nearest to the flame to the other end.</p>	Student listens carefully.	Lecture Method Demonstration Method.	
Activity		<p>Teacher's Statement</p> <p>The process by which heat is transferred from the hotter end to the colder end of an object is known as conduction. In solids, generally, the heat is transferred by the process of conduction.</p> <p>When energy is transferred to an object, the energy of the particles increases. This means the particles have more kinetic energy and start to move and vibrate faster. As the particles are moving faster, they "bump" into other particles and transfer some of their energy to the neighbouring particles. In this way, the energy is transferred through the substance to the other end.</p>	They start to fall. The end nearest to the flame falls first.	<p>Class: VII Subject: Science Topic: Transfer of Heat <u>CONDUCTION</u></p>	Student listens carefully and notes down in their notebook.
Conduction	Student explains the process of conduction.			<p>○○○○ ○○○○ ○○○○ ○○○○ →○○○○ →○○○○ ○○○○ ○○○○ ○○○○ ↑ ↑ ↑ HEAT HEAT HEAT</p> <p><u>PROCESS OF CONDUCTION</u></p>	explain the process of conduction.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
Convection	Student describes the process of convection through an activity.	<p>Pupil teacher does the following activity :-</p> <ul style="list-style-type: none"> A beaker is half filled with cold water. A small amount of potassium permanganate is added to the beaker. Do not stir. Heat the water directly using the side of the beaker. <p>Ques: What did you see as the water started to warm up in the beaker?</p> <p>Ques: What is happening to the potassium permanganate in the beaker?</p> <p>As the liquid is heated, the water expands!</p> <p>This is because the particles in liquids & gases gain kinetic energy when they are heated and start to move faster.</p> <p>Convection is the process by which heat is transferred through liquids and gases from one place to another by the actual movement of matter.</p>	<p>Student listens carefully.</p> <p>The purple colour of the dissolved potassium permanganate moves in a circle upwards through the water.</p> <p>Potassium permanganate dissolves in the water.</p>	<p>Lecture Method.</p> <p>Question-Answer Method.</p> <p>Demonstration Method.</p>	<p>Describe the process of convection through an activity.</p>
Radiation	Student defines radiation	<p>Teacher's Statement</p> <p>Radiation means energy transfer from one medium to another without affecting the intermediate medium.</p> <p>It doesn't require objects to be in contact with each other or the movement of particles.</p>	<p>Student listens carefully and writes it in their notebook.</p>	<p>Lecture Method.</p>	<p>Define Radiation</p>



Closure

So students, today we discussed about

- 3 ways of transfer of heat from one object to another.
- Conduction
- Convection
- Radiation

Ques: State the difference between conduction, convection and radiation.

Ques: Fill in the blanks:-

- i) No medium is required for transfer of heat by process of _____.
- ii) A cold steel spoon is dipped in a cup of hot milk. It is transferred to its other end by the process of _____.

Home Assignment

Ques: Why wearing more layers of clothing during winter keeps us warmer than wearing just one thick piece of clothing?

Signature of the Intern

Assessment of Lesson

Name of Intern: Julie Koshy Date 16/01/2020
Topic: Transfer of Heat Lesson Plan No. 4

Category Under Supervision		Rating Scale				
		Not Used	Poor	Avg.	Good	Very Good
Lesson Plan Introduction	1. Objectives stated in behavioural terms	0	1	2	3	4
	2. Student's previous knowledge tested	0	1	2	3	4
	3. Topic was announced effectively	0	1	2	3	4
Presentation	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
	9. Skill of Illustration with examples	0	1	2	3	4
Use of Teaching Aid	10. Adequate & appropriate	0	1	2	3	4
	11. Quality of teaching aid	0	1	2	3	4
Chalk Board Work	12. Organized and adequate	0	1	2	3	4
	13. Appropriateness and clarity	0	1	2	3	4
Management of the Class	14. Class environment & discipline	0	1	2	3	4
	15. Supervision of student's work	0	1	2	3	4
Evaluation	16. Skill of Closure	0	1	2	3	4
	17. Based on the behavioral objectives	0	1	2	3	4
	18. Quality of home assignment	0	1	2	3	4

For all comments and suggestions:-

(Class Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of lesson with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Galgotias University Uttar Pradesh, G.B. Nagar
School of Education

Lesson Plan No. 5

Date 16/01/2020

Name of the Internship School National Victor Public School

Class & Section VIII - D Period 2nd Duration 40 mins

Subject Science Sub-Subject Chemistry

Topic Synthetic Fibres

Specific Objectives After my class, students will be able to

- define synthetic fibres
- write the types of synthetic fibres.
- state the uses of synthetic fibres

Previous Knowledge

Students have basic knowledge of fibres.

Introduction

Ques : What are the clothes we wear made of?

Ans : Fabrics

Ques : What are fabrics made of?

Ans : Fibres

Ques : Name any 2 natural fibres.

Ans : Silk, wool.





Ques : Rayon comes under which type of fibre?

Ans : No Answer.



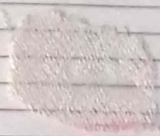
Announcement of the Topic

"Today, we'll discuss about "Synthetic Fibres".

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Fibre	Student states the two types of fibres.	<u>Teacher's Statement</u> A very thin thread like substance which is used to make clothes is called fibre. <u>Types of fibres</u> i) Natural fibres: Fibres which are obtained from natural sources are called natural fibres. These are obtained from plants or animals and they can be easily spun into thread, which in turn can be woven, matted or knitted. eg: wool, jute, cotton, silk etc. ii) Synthetic fibres: Fibres which are formulated artificially by human beings are called synthetic fibres. It is formulated by joining together a chain of small units that are actually chemical groups to form huge units called polymer. eg: Rayon, Polyester, Nylon, acrylic etc.	Student listens carefully and writes down the different types of fibres in their notebook.	Lecture Method. Demonstration Method.  - Nylon  - Silk  - Polyester  - Nylon	State the two types of fibres.
2. Classification of Synthetic Fibres	Student classifies the different types of synthetic fibres.	<u>Teacher's Statement</u> Synthetic fibres can be classified into :- i) Rayon: - Artificial silk, which is moisture absorbent, comfortable to wear, soft on skin with restrained dyeing.	Student listens carefully.	Lecture Method.	

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
		<p>ii) Nylon :- This fibre was first made from coal, water & air.</p> <ul style="list-style-type: none"> - Good elasticity and strength. - Used in making ropes, belts, sleeping bags, tents. 		 <p>- Nylon</p>	
		<p>iii) Polyester :- A durable fibre possessing the characteristic resiliency due to which it is more resistant to wrinkles.</p> <ul style="list-style-type: none"> - They are also impervious to chemicals, shrinking & stretching. - Used in making shirts, jackets, curtains, sarees. 		 <p>- Polyester</p>	
		<p>iv) Acrylic :- It is formulated from a polymer called polyacrylonitrile.</p> <ul style="list-style-type: none"> - It is light weight, soft, and has resemblance with wool. - Impervious to moths, oils & chemical substances. - Used in making sweaters, suits, gloves & carpets. 		 <p>- Acrylic</p>	Classify the different types of synthetic fibres.
3. Characteristics of Synthetic fibres.	Student lists the characteristics of synthetic fibres.	<p>Teacher's Statement</p> <p>Synthetic fibres possess the following characteristics :-</p> <ol style="list-style-type: none"> Durable & impervious. Light weight, smooth & soft. Resilient to wrinkles. Can be easily washed & dried. Their tenacity makes them useful in many industries. 	Student notes down the characteristics of synthetic fibres in their notebook.	Lecture Method.	List the characteristics of synthetic fibres.

Closure

So students, today we discussed about

- Fibres

- Synthetic fibres - Classification & characteristics

Ques:- Explain why some fibres are called synthetic

Ques:- Fill in the blanks:-

a) Synthetic fibres were also called _____ fibres.

b) Synthetic fibres are synthesised from man-made material called _____.

c) Rayon is different from synthetic fibres because it has _____ appearance.

Home Assignment

Ques: Give examples which indicate that nylon fibres are very strong.

Signature of the M

Assessment of Lesson

Name of Intern: Julie Koshy Date: 16/01/2020
 Topic: Synthetic Fibres Lesson Plan No: 5

Category Under Supervision	Rating Scale					
	Not Used	Poor	Avg.	Good	Very Good	
Lesson Plan	0	1	2	3	4	
Introduction	1. Objectives stated in behavioural terms	0	1	2	3	4
	2. Student's previous knowledge tested	0	1	2	3	4
	3. Topic was announced effectively	0	1	2	3	4
Presentation	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
	9. Skill of Illustration with examples	0	1	2	3	4
	10. Adequate & appropriate	0	1	2	3	4
	11. Quality of teaching aid	0	1	2	3	4
	12. Organized and adequate	0	1	2	3	4
Use of Teaching Aid	0	1	2	3	4	
Chalk Board Work	13. Appropriateness and clarity	0	1	2	3	4
	14. Class environment & discipline	0	1	2	3	4
Management of the Class	15. Supervision of student's work	0	1	2	3	4
Evaluation	16. Skill of Closure	0	1	2	3	4
	17. Based on the behavioral objectives	0	1	2	3	4
	18. Quality of home assignment	0	1	2	3	4

For all comments and suggestions:-
 Press Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of lesson with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Galgotias University Uttar Pradesh, G.B. Nagar
School of Education

Lesson Plan No. 6

Date 20-01-2020

Name of the Internship School National Victor Public School

Class & Section VII-A Period 5th Duration 40 min

Subject Science Sub-Subject Chemistry

Topic Physical and Chemical Changes

Specific Objectives After my class, students will be able to

- differentiate between physical and chemical change.
- compare the properties of a substance before and after a chemical change.
- identify the physical & chemical changes.

Previous Knowledge

Students have good knowledge about different changes a substance undergoes.

Introduction

Ques : What happens when ice melts?

Ans : It turns into water.

Ques : What happens if we heat water?

Ans : It turns into vapour.

Ques : What does ice & water undergo?

Ans : A change.

Ques : What is this type of change referred to as?

Ans : No Response.

Announcement of the Topic

Today, we'll discuss about "Physical & Chemical Changes".

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Change	Student cites different ways a given substance may undergo a change.	<p><u>Teacher's Statement</u></p> <p>Change is a universal phenomenon. Change that occurs in a substance may change their :-</p> <p>i) Colour (Raw mango - green in colour turns yellow)</p> <p>ii) Shape, size (sapling turning into a plant)</p> <p>iii) Hardness (Raw rice is hard but turns soft when it is cooked)</p> <p>iv) Composition (Burning of coal)</p>	Student listens carefully and notes down in their notebook.	Lecture Method.	Write the different ways a given substance may undergo a change.
2. Physical Change	Student explains physical change with examples.	<p><u>Teacher's Statement</u></p> <p>* Properties such as shape, size, colour and state of a substance are called physical properties.</p> <p>* A change in which a substance undergoes a change in its physical properties is called <u>Physical Change</u>.</p> <p>* It is a <u>Temporary change</u> in which no new substance is formed and the chemical composition remains the same.</p> <p>eg: Melting of ice, sublimation, making of sugar solution, cutting of vegetables etc.</p> <p><u>Ques:</u> What happens when a candle is lighted?</p> <p><u>Ans:</u> What changes does it undergo? It undergoes both physical & chemical changes.</p>	Student listens carefully and notes down in their notebook.	Lecture Method Demonstration Method.	Candle melts. It's a physical change. Explain Physical change with examples.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
3. Chemical Change	Student describes chemical changes with examples.	<u>Teacher's Statement</u> Changes which results in the formation of new substances with completely new chemical and physical properties composition is called a chemical change. These changes are irreversible. Eg: Photosynthesis - $6CO_2 + 6H_2O \xrightarrow[\text{chlorophyll}]{\text{sunlight}} C_6H_{12}O_6 + 6O_2$	Student listens carefully and notes down in their notebook.	Lecture Method	
o Meaning		- Respiration - $C_6H_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O + \text{energy}$			
o Examples		- Combustion $CH_4 + O_2 \rightarrow CO_2 + H_2O$ - Spoilage of food, cutting a slice of apple etc.			Describe a chemical change with examples.
o Characteristics	Student states the characteristics of a chemical change	<u>Teacher's Statement</u> <u>Characteristics of a Chemical Change</u> i) evolution of gas : $CaCO_3 \xrightarrow{\Delta} CaO + CO_2 \uparrow$ ii) Change of colour : $Fe + CuSO_4 \rightarrow FeSO_4 + Cu$ (blue solution) (green) iii) Formation of precipitate : $Na_2SO_4 + BaCl_2 \rightarrow BaSO_4 \downarrow + NaCl$ iv) Production of sound : $CH_3COOH + NaHCO_3 \rightarrow CH_3COONa + CO_2 + H_2O$ (Vinegar) (Baking Soda) v) Change of smell.	Student listens carefully and notes down the characteristics of a chemical change in their notebook.	Lecture Method	State the characteristics of a chemical change.

Closure

So student, today we discussed about

- Physical Change
- Chemical Change

Ques: Fill in the blanks :-

1. Cutting paper is a — change.
2. Burning fire is an example of — change.
3. Boiling of water is a — change.
4. Rusting of iron is a — change.

Home Assignment

Ques: Find 3 examples of chemical & physical changes in everyday life. Explain why each of these changes is either chemical/physical.

Signature of the Ment

Assessment of Lesson

Name of Intern: Julie Kochy Date: 20-01-2020
 Topic: Physical and Chemical Changes Lesson Plan No. 6

Category Under Supervision	Rating Scale					
	Not Used	Poor	Avg.	Good	Very Good	
A. Lesson Plan	1. Objectives stated in behavioural terms	0	1	2	3	4
	2. Student's previous knowledge tested	0	1	2	3	4
	3. Topic was announced effectively	0	1	2	3	4
B. Introduction	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
	9. Skill of Illustration with examples	0	1	2	3	4
C. Presentation	10. Adequate & appropriate	0	1	2	3	4
	11. Quality of teaching aid	0	1	2	3	4
D. Use of Teaching Aid	12. Organized and adequate	0	1	2	3	4
	13. Appropriateness and clarity	0	1	2	3	4
E. Chalk Board Work	14. Class environment & discipline	0	1	2	3	4
	15. Supervision of student's work	0	1	2	3	4
F. Management of the Class	16. Skill of Closure	0	1	2	3	4
	17. Based on the behavioral objectives	0	1	2	3	4
G. Evaluation	18. Quality of home assignment	0	1	2	3	4

Over all comments and suggestions:-

(Press Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of lesson with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Galgotias University Uttar Pradesh, G.B. Nagar
School of Education

Lesson Plan No. 7

Date 20.01.2020

Name of the Internship School National Victor Public School

Class & Section VI-B Period 2nd Duration 40mins

Subject Science Sub-Subject Physics

Topic Light, Shadows and Reflections

Specific Objectives After my class, students will be able to

- define luminous & non-luminous objects.
- differentiate between transparent, translucent and opaque substances.
- explain how a shadow is formed.

Previous Knowledge

Students have general idea about different bodies which emit light.

Introduction

Ques : What is light?

Ans : Form of energy.

Ques : How does it travel?

Ans : Straight line.

Ques : Bodies which emit light are referred to as _____.

Ans : No Response.

Announcement of the Topic

"Today, we will discuss about "Light, Shadows and Reflections."

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Luminous and Non-luminous Bodies	Student defines luminous and non-luminous bodies with some examples	Pupil Teacher makes the student do the following Activity :- Activity :- Take a rectangular box (shoe box) and cut a peeping window on one of its smaller faces. Place few objects such as pens, pencils, geometry box, diaries etc. inside the box. Cover the box as well as your head with a piece of thick black cloth. Peep into the box through the window. Can you see the things placed inside the box? No. Now place a lighted torch inside the box and peep into the box again. Can you see the things placed inside the box? Yes.	Student listens carefully and follows the instructions provided.	Lecture Method Demonstration Method	
o Activity				Boardswork Class: VI Subject: Science Topic: LIGHT, SHADOWS AND REFLECTIONS	
o Definition			o Not able to see anything.	<u>Luminous Bodies</u> ↓ emit light (the sun, stars, burning candle)	
o Examples				<u>Non-luminous bodies</u> do not emit light (tables, chairs, the earth)	
		Therefore, it can be concluded that we see objects that do not emit light of their own only when light from some other source is made to fall on them.	o Student notes down the definition in their notebook.		
		Bodies which emit (give out) light are called luminous bodies. eg: The sun, stars, burning candle, hot filament in an electric bulb.			
		Bodies which do not emit light are called non-luminous objects. eg: Tables, chairs, the earth etc.			Define luminous and non-luminous bodies with some examples

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
2. Transparent, Translucent and Opaque Substances	Student differentiates between transparent, translucent and opaque substances.	<p>Teacher's Statement</p> <p>When a ray of light falls on an object, either of the following will occur :-</p> <ul style="list-style-type: none"> The light may pass through the object completely. Only a part of the light may pass through the object or The light may not pass through the object at all. <p>Why do different objects behave differently to light?</p> <p>This is because different objects are made up of different materials and different materials show different behaviours towards light.</p> <p>Based on this, materials can be classified into 3 categories :-</p> <ol style="list-style-type: none"> Transparent : A substance which allows light to pass through it is called transparent substance. eg: glass, water etc. Opaque : A substance which does not allow light to pass through it is called an opaque substance. eg: Metals, wood etc. Translucent : A substance which allows only a part of light to pass through it is called a translucent substance. eg: skin, Wax-paper, etc. 	<p>Student listens carefully.</p> <p>Response.</p> <p>Student listens carefully and notes down the difference between transparent, opaque and translucent bodies in their notebook.</p>	<p>Lecture Method.</p> <p>Illustration with examples</p> <p>Boardwork</p> <p>Class: VI Subject: Science Topic: LIGHT, SHADOWS AND REFLECTIONS</p> <p>Transparent - Allows light to pass Opaque - Doesn't allow light to pass Translucent - Allows only a part of light to pass through it.</p>	<p>Differentiate between transparent, opaque and translucent substances.</p>

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
3. Shadows	Student explains how a shadow is formed.	<p><u>Teacher's Statement</u></p> <p>Did you observe your shadow while walking in the sun?</p> <p>Your shadow moves with you.</p> <p>* Your shadow is very long in the morning when the sun rises.</p> <p>Its length goes on decreasing till noon and is shortest at the noon time.</p> <p>The shadow goes on increasing in the aft. noon and becomes very long just before the sun sets.</p> <p>* You cannot see shadows in dark.</p> <p>* Shadow is formed when light rays made to fall on an opaque object.</p> <p>* Shadow of an object gives an idea about its shape.</p> <p>* Shadow is formed only if there is a surface or a screen behind the object.</p>	Student listens carefully and notes the important points in their note book.	<p>Lecture Method.</p> <p>boardwork</p> <p>Class: VI Subject: Science Topic: LIGHT, SHADOWS AND REFLECTIONS</p> <div style="border: 1px solid black; padding: 5px; margin: 5px 0;"> <p><u>Shadow</u></p> <p>↓</p> <ul style="list-style-type: none"> • formed when light is made to fall on an opaque object • formed only if there is a surface or a screen behind the object. </div>	Explain how a shadow is formed.

Closure

So students, today we discussed about

- Luminous and Non-luminous bodies
- Transparent, Opaque and Translucent substances
- Formation of shadow.

Ques: Give an example of translucent substance.

Ques: In which direction does the shadow of an object fall?

Ques: Where should the screen be placed to get the shadow of an object?

Home Assignment

Ques: Shadow of an object may be dark and clear or fuzzy.

What makes the shadow dark & clear or fuzzy?

Assessment of Lesson

Name of Intern: Julie Koshy Date: 20/01/2020
 Topic: Light, Shadows and Reflections Lesson Plan No: 7

Category Under Supervision	Not Used	Rating Scale				
		Poor	Avg.	Good	Very Good	
Lesson Plan Introduction	1. Objectives stated in behavioural terms	0	1	2	3	4
	2. Student's previous knowledge tested	0	1	2	3	4
	3. Topic was announced effectively	0	1	2	3	4
Presentation	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
	9. Skill of Illustration with examples	0	1	2	3	4
Use of Teaching Aid	10. Adequate & appropriate	0	1	2	3	4
	11. Quality of teaching aid	0	1	2	3	4
Chalk Board Work	12. Organized and adequate	0	1	2	3	4
	13. Appropriateness and clarity	0	1	2	3	4
Management of the Class	14. Class environment & discipline	0	1	2	3	4
	15. Supervision of student's work	0	1	2	3	4
Evaluation	16. Skill of Closure	0	1	2	3	4
	17. Based on the behavioral objectives	0	1	2	3	4
	18. Quality of home assignment	0	1	2	3	4

For all comments and suggestions:-

(Please mention Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of lesson with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Signature of the Mentor

Lesson Plan No. 8

Date 21-01-2020

Name of the Internship School National Victor Public School

Class & Section VI-B Period 2nd Duration 40mins

Subject Science Sub-Subject Physics

Topic Light, Shadows and Reflection

Specific Objectives After my class, students will be able to

- explain why light travels in a straight line.
- cite the advantages of a pinhole camera.
- define reflection of light.

Previous Knowledge

Students have good knowledge about luminous and non-luminous bodies and transparent, opaque & translucent substances.

Introduction

Ques: What are luminous objects?

Ans: Objects which emit light are referred as luminous objects.

Ques: Give 2 examples of luminous bodies.

Ans: The sun, stars etc.

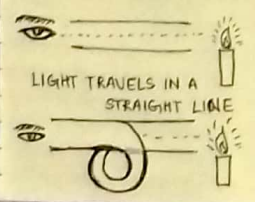
Ques: "The process of sending back of light rays which falls on the surface of an object is called _____."

Ans: No Response.

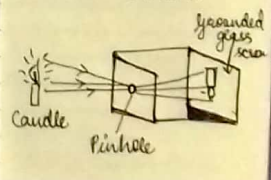
Announcement of the Topic

"Today, we will discuss about 'Light, Shadows and Reflections'."

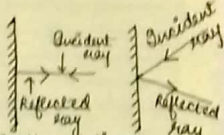
Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Light travels in a straight line.	Student explains why light travels in a straight line with the help of an activity.	<p>Pupil Teacher demonstrates the following activity in the classroom.</p> <p><u>Activity :-</u></p> <ul style="list-style-type: none"> o Ties a lighted candle on the table. o Then stretches a narrow tube straight and then asks two of the students to look through it at the flame. <p>Do you see the light coming from the flame?</p> <ul style="list-style-type: none"> o Now coil the rubber tube and look through it at the flame. <p>Do you see the light coming from the flame?</p> <p>Thus, this experiment shows light travels in a straight line.</p>	<p>Student listens carefully and understands the concept that light travels in a straight line.</p> <p>Yes, the light coming from the flame is visible.</p> <p>No, the light coming from the flame is not seen.</p>	<p>Lecture Method.</p> <p>Demonstration Method.</p> <p>Class: II Subject: Science Topic: LIGHT, SHADOWS AND REFLECTIONS</p>  <p>The diagram consists of two parts. The top part shows an eye on the left and a lit candle on the right. A dashed line representing light travels in a straight line from the candle through a narrow tube to the eye. Below this, the text 'LIGHT TRAVELS IN A STRAIGHT LINE' is written. The bottom part shows the same eye and candle, but the tube between them is curved. The dashed line representing light is blocked by the curve of the tube and does not reach the eye.</p>	<p>Explain light travels in a straight line through an activity.</p>
2. Pinhole camera	Student describes the construction of a pinhole camera	<p><u>Teacher's Statement</u></p> <p>Pinhole camera is based on principle that light travels in a straight line.</p> <p><u>Construction</u></p> <p>A pinhole camera consists of a cardboard or wooden box with a pinhole in its front face and a translucent ground glass or white paper screen at the back.</p>	<p>Student listens carefully.</p>	<p>Lecture Method</p> <p>Demonstration Method.</p>	

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
		<p>The distance between the pinhole and the screen may be fixed or adjustable.</p> <p>The distance between the pinhole and the screen can be adjusted by moving the screen towards or away from the pinhole.</p>	<p>Student listens carefully</p>		
o Working		<p>When the pinhole of the camera is turned towards a bright object, a real, inverted image of the object is formed on the screen.</p>			<p>Describe the construction of a pinhole camera.</p>
o Advantages	<p>Student cites the advantages of a pinhole camera</p>	<p><u>Teacher's Statement</u></p> <p>A pinhole camera has the following advantages:-</p> <ol style="list-style-type: none"> No focussing required. There is no lens in pinhole camera. <p>Therefore, the image is free from spherical and chromatic aberrations.</p>	<p>Student listens carefully and notes down the advantages and disadvantages in their notebook.</p>	<p>Lecture Method</p> <p>Class: VI Subject: Science Topic: LIGHT, SHADOWS AND REFLECTIONS</p> <p>PINHOLE CAMERA</p> 	<p>Cite the advantages of a pin hole camera.</p>
o Disadvantages	<p>Student enlists the disadvantages of a pinhole camera</p>	<p>A pinhole camera has the following disadvantages:-</p> <ol style="list-style-type: none"> The image formed doesn't give any details. Image is obtained on the screen. No permanent record of the image can be obtained. It can't be used for studying moving objects. 			<p>Enlist the disadvantages of a pinhole camera.</p>

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
3. Mirrors & Reflection	Student defines the phenomenon of reflection.	<p><u>Teacher's Statement</u></p> <p><u>Mirror</u> :- A smooth, highly polished reflecting surface is called mirror. eg: Plane Mirror</p> <p><u>Reflection</u> :- The phenomenon in which a ray of light after falling on a smooth polished surface or a mirror returns to the same medium is called reflection.</p> <p>* The ray of light that falls on a mirror is called <u>Incident ray of light</u>.</p> <p>* The ray of light that comes back after suffering reflection from a mirror is called <u>Reflected ray of light</u>.</p>	Student listens carefully and notes down in their notebook	<p>Lecture Method</p> <p><u>Boardwork</u></p> <p>Class: VI Subject: Science Topic: LIGHT, SHADOWS AND REFLECTIONS</p>	Define the phenomenon of reflection
				 <p>Reflection of light rays by a plane mirror</p>	

Closure

- So students, today we discussed about
- light travels in a straight line
 - Pinhole camera - Advantages & Disadvantages
 - Mirrors & Reflection

Ques: Name a device based on the rectilinear propagation of light

Ques: Show the reflection of light by a plane mirror

Home Assignment

Ques: Differentiate between the image and the shadow of an object.

Signature of the Intern

Assessment of Lesson

of Intern: Julie Koshy Date: 21/01/2020
 LIGHT, SHADOWS AND REFLECTIONS Lesson Plan No. 8

Category Under Supervision	Rating Scale				
	Not Used	Poor	Avg.	Good	Very Good
Lesson Plan	0	1	2	3	4
Introduction	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
Presentation	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
Use of Teaching Aid	0	1	2	3	4
	0	1	2	3	4
Chalk Board Work	0	1	2	3	4
	0	1	2	3	4
Management of the Class	0	1	2	3	4
	0	1	2	3	4
Evaluation	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4

All comments and suggestions:-
 Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of work with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Lesson Plan No. 9

Date 22.01.2020

Name of the Internship School National Victor Public School

Class & Section VI-A Period 2nd Duration 40 mins

Subject Science Sub-Subject Chemistry

Topic Separation of Substances

Specific Objectives After my class, student will be able to

- classify mixtures into 2 - Homogeneous & Heterogeneous mixture.
- differentiate between pure & impure substances.
- state the need for separation of constituents of a mixture

Previous Knowledge

Students have good knowledge about substances and mixtures.

Introduction

We all have seen our parents & grandparents separating small stones or pebbles from the rice grains, pulses, filtering tea leaves before serving tea and lot more.

So, do you think separation of substances is required?

No Response.

Announcement of the Topic

"Today, we'll discuss about "Separation of Substances".

Presentation

Teaching Point	Behavioural Objectives	Teacher's Statement	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Mixtures	Student explains the term 'mixtures' with suitable examples	<p><u>Teacher's Statement</u></p> <p>Materials which contain more substances in any proportion are called mixtures.</p> <ul style="list-style-type: none"> • A mixture may be a solid, or a gas. • The various substances present in a mixture are called components or constituents. <p>For eg: in a mixture of sugar and sand, sugar & sand are called the constituents of this mixture.</p>	Student listens carefully and notes down in their notebook	Lecture Method Demonstration Method	Explain the term 'mixtures' with suitable examples
2. Classification of Mixtures	Student differentiates between heterogeneous and homogeneous mixtures	<p><u>Teacher's Statement</u></p> <p>All mixtures can be divided into 2 categories - Heterogeneous & Homogeneous</p> <p><u>Heterogeneous Mixtures</u></p> <p>Mixtures in which particles of substances present can be seen easily are called heterogeneous mixtures.</p> <p>eg: Water + Oil Chalk in water Heap of wheat (husk & grains) Dust in air Soil (clay, salts, dead & dead plants)</p>	Student listens carefully and notes down the examples	Lecture Method Demonstration Method	

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
Homogeneous Mixtures		<p><u>Homogeneous Mixtures</u></p> <p>Mixtures in which the particles of the substances present cannot be seen are called homogeneous mixtures</p> <p>eg: • Solutions of sugar & salt water.</p> <ul style="list-style-type: none"> • Cold drinks and syrups etc - mixture of sugar, colouring agent in water • Air (gases - nitrogen, oxygen, CO₂) 			Differentiate between Homogeneous & Heterogeneous mixtures.
Pure and Impure substances	Student differentiate between pure and impure substances	<p><u>Pure and Impure Substances</u></p> <p><u>Pure Substance</u> A single substance of a definite composition is called a pure substance.</p> <p>eg: pure iron, pure copper, pure gold, pure oxygen</p> <ul style="list-style-type: none"> • A pure substance contains particles of only one type. • A pure substance has a fixed melting or fixed boiling point. • A pure substance has its characteristic taste, colour and odour. <p><u>Impure Substance</u> A substance in which some other substances are also present in smaller amounts called an impure substance.</p>	Student listens carefully and notes down the difference between pure & impure substances in their notebook.	<p>Lecture Method.</p> <p><u>Boardwork</u></p> <p>Class: VI Subject: Science Topic: SEPARATION OF SUBSTANCES</p> <p><u>Pure Substance</u> eg: pure iron, pure gold, pure oxygen etc.</p> <p><u>Impure Substance</u> eg: Dust in air, soil etc</p>	Differentiate between pure & impure substance

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
4. Purpose of separating the constituents of a mixture	Student states the need for separating the constituents of a mixture.	<p><u>Teacher's Statement</u></p> <p>The separation of constituents of a substance is carried out for the following purposes:</p> <p>i) To remove any harmful or undesirable constituent. eg: Small pieces of stones and undesirable seed grains are removed from rice and dal before cooking.</p> <p>ii) To obtain a pure sample of a substance. Impurities present in any substance are removed to obtain a pure sample of the substance. Pure substances are needed in medicines, research, industry and also in daily use.</p> <p>iii) To obtain useful substance. A mixture may contain one or more useful constituents. eg: Petrol, Diesel, Kerosene etc. are obtained from crude petroleum by a method called fractional distillation.</p>	Student listens carefully and writes down in her notebook.	<p>Lecture Method</p> <p>Boardwork</p> <p>Class: VI Subject: Science Topic: SEPARATION OF SUBSTANCES</p> <p>Need:</p> <ul style="list-style-type: none"> To remove any harmful or undesirable constituent To obtain a pure sample of a substance To obtain useful substances. 	State the need for separating the constituents of a mixture.

Closure

So students, today we discussed about

- Homogeneous & Heterogeneous Mixtures
- Pure & Impure Substances
- need for separation of constituents of a mixture

Ques: What type of mixture is a heap of wheat?

Ques: Differentiate between homogeneous & heterogeneous mixtures

Ques: Name a gaseous mixture containing oxygen

Home Assignment

Ques: A sample of water boils at 102°C and has a salty taste.

What do you conclude from this?

Signature of the Intern

Assessment of Lesson

of Intern: Julie Koshy
Separation of Substances

Date 22/1/2020

Lesson Plan No. 9

Category Under Supervision	Rating Scale				
	Not Used	Poor	Avg.	Good	Very Good
Lesson Plan	0	1	2	3	4
Introduction	0	1	2	3	4
Presentation	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
Use of Teaching Aid	0	1	2	3	4
Chalk Board Work	0	1	2	3	4
	0	1	2	3	4
Management of the Class	0	1	2	3	4
	0	1	2	3	4
Evaluation	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4

All comments and suggestions:-

Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of (with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Lesson Plan No. 10

Date 24.01.2020

Name of the Internship School National Victor Public School

Class & Section VI-A Period 2nd Duration 40 mins

Subject Science Sub-Subject Chemistry

Topic Separation of Substances (Solid from other Solids)

Specific Objectives After my class, students will be able to

- describe the method of handpicking.
- explain the process of sieving.
- describe the method of winnowing.
- cite examples for different methods.

Previous Knowledge

Students have general idea about the need of separating the constituents of a mixture.

Introduction

Ques: What do you mean by 'separation of substances'?

Ans: Process of separating one or more components from a mixture.

Ques: Why is separation of substances required?

Ans: To remove any harmful or undesirable component.

Ques: List different methods for separating substances from their mixture.

Ans: No Response.

Announcement of the Topic

Today, we'll discuss about "Separating a solid from other solids".

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Separating a solid from other solids	Student describes the method used for separating the desired solid component from a mixture of solids	<p><u>Teacher's statement</u></p> <p>The methods described below are used for separating desired solid component a mixture of solids.</p> <p><u>Handpicking</u>: This method is useful only when -</p> <ol style="list-style-type: none"> the unwanted material is present in smaller quantity the shape, size & colour of the unwanted material is different from that of the useful one. <p>Most foodgrains such as, rice, pulses etc contain very small pieces of stones and sometimes insects, also. These are removed by hand picking.</p>	Student listens carefully and notes down in their notebook.	Lecture Method Demonstration Method.	
o Handpicking					Describe the method of handpicking.
o Sieving	Student elaborates the method of sieving.	<p><u>Sieving</u> :- This method is used for separating solid constituents of a mixture which differ in their sizes.</p> <p>In this method, a sieve having holes of proper size is used. The bigger particles are retained by the sieve, whereas the smaller ones pass through it.</p> <p>This method is used for :-</p> <ul style="list-style-type: none"> * separating chaff from wheat * separating fine sand from gravel * separating cashewnuts of different sizes * separating pearls of different sizes. 	Student listens carefully and writes down the methods for separation of solid from other solids in their notebooks.	Lecture Method Deductive Method.	elaborate the method of sieving.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
o Winnowing	Student explains the method of winnowing.	<p><u>Winnowing</u> :-</p> <p>This method is based on the property that grains are heavier than husk and chaff.</p> <p>Farmers use this method to separate wheat grains from husk and chaff. They drop the mixture of wheat grains, husk and chaff from a height. The lighter husk and chaff are carried by the wind and fall at a small distance away. The wheat grains, being heavier, fall almost vertically to form a separate heap nearby.</p>	Student listens carefully and notes down in their notebook.	Lecture Method Illustration with examples	Explain the method of winnowing.
o Threshing	Student describes the method of threshing.	<p><u>Threshing</u> :-</p> <p>The process of separating grain from sun-dried stalks is called threshing. In this process, the stalks are broken against any hard surface such as a large rock or a stone slab etc. to free the grains.</p> <p>Threshing is also done with the help of animals. Large quantities of stalks can be subjected to a machine called a threshing machine to obtain grains more easily.</p>	Student listens carefully and notes down in their notebook.	Lecture Method	Describe the method of threshing.

Closure

So students, today we discussed about the different methods for separating the desired solid component from a mixture of solids.

- Handpicking
- Sieving
- Winnowing
- Threshing

Ques :- Which method is used for separating seeds of stones & insects from foodgrains?

Ques :- Name the machine used for separating grain from sun-dried stalks.

Home Assignment

Ques :- Describe the methods used for separating a solid from other solids.

Signature of the Intern

Assessment of Lesson

Signature of the Intern: Julie Koshy Date: 24/11/2020
 Topic: Separation of Substances Lesson Plan No: 10

Category Under Supervision	Rating Scale				
	Not Used	Poor	Avg.	Good	Very Good
Lesson Plan	0	1	2	3	4
Introduction	0	1	2	3	4
Presentation	0	1	2	3	4
Use of Teaching Aid	0	1	2	3	4
Chalk Board Work	0	1	2	3	4
Management of the Class	0	1	2	3	4
Evaluation	0	1	2	3	4

All comments and suggestions:-

Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of (with the help of Students)

Julie
 Signature of the Intern

Signature of the Mentor

Lesson Plan No. 11

Date 24-01-2020

Name of the Internship School National Victor Public School

Class & Section VI-A Period 4th Duration 40mins

Subject Science Sub-Subject Chemistry

Topic Separating Insoluble Solids from Liquids

Specific Objectives After my class, students will be able to

- explain the process of sedimentation & decantation.
- describe the process of loading.
- illustrate the process of filtration.

Previous Knowledge

Students have good knowledge of separation of substances.

Introduction

Ques: What do you mean by 'separation of substances'?

Ans: Process of separating one or more components from a mixture.

Ques: List some methods of separating a solid from other solids.

Ans: Handpicking, Winnowing, Sieving etc.

Ques: How can we separate insoluble solids from liquids?

Ans: No Response.

Announcement of the Topic

"Today, we'll discuss about 'Separating Insoluble Solids from liquids'".

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Separating insoluble solids from liquids	Student explains the methods used for separating insoluble solids from liquids.	<p><u>Teacher's Statement</u></p> <p>The following methods are used for separating solids from liquids.</p> <p><u>Sedimentation and Decantation</u></p> <p>Sedimentation and decantation are generally used for separating coarse particles of a solid from a liquid.</p> <p>eg:- The coarse particles of sand in muddy water can be separated by decantation.</p> <p>Before cooking, rice, pulses are washed with water. The water is removed by decantation.</p> <p><u>Activity:</u> (Pupil teacher instructs the student to carry out the following:-)</p> <ul style="list-style-type: none"> Take small quantity of muddy water (mixture of sand & water) in a beaker. Allow it to stand for some time. What do you observe? This settling down of heavy particles is called sedimentation. Transfer the clear supernatant (upper layer of water) carefully into another beaker. <p>This process of separating the liquid without disturbing the settled solid particles - Decantation.</p>	<p>Student listens carefully and writes the important points in their notebooks.</p> <p>Student follows the instructions.</p> <p>particles of sand settle down to the bottom of the beaker.</p>	<p>Lecture Method</p> <p>Demonstration Method</p> <div data-bbox="997 801 1268 1131" data-label="Diagram"> </div>	<p>Explain the methods used for separating insoluble solids from liquids.</p>

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
2. Loading	Student explains loading.	<p><u>Teacher's Statement</u></p> <p>During sedimentation, the coarse particles settle down quickly. Fine particles of clay settle very slowly. The fine particles can be made to settle faster by dissolving a small quantity of alum in muddy water. This method is called loading.</p> <p>When alum is added to muddy water, the finer particles get loaded with alum particles and form larger aggregates. This makes them to settle faster.</p>	Student listens carefully and notes down in their notebook	<p>Lecture Method</p> <p>Demonstration Method</p> <div data-bbox="997 779 1289 1124" data-label="Diagram"> </div>	explain loading.
3. Filtration	Student describes the process of filtration	<p><u>Teacher's Statement</u></p> <p>This method is used for separating fine insoluble solid particles from the liquid. In this method, the mixture is allowed to pass through a filter. The insoluble solid particles are retained by the filter and the clear liquid is collected as filtrate. eg: Filtering tea by using a strainer. Here, the tea leaves are retained by the strainer.</p>	Student listens carefully and notes down in their notebook	Lecture-cum - Demonstration Method.	Describe the process of filtration.

Closure

So students, today we discussed about separation of insoluble solids from liquids by the process of Sedimentation & Decantation.

- Loading
- Filtration

Ques: Name the method used for obtaining clear water from muddy water.

Ques: Which method is best for separating sand from water - Sedimentation, decantation or filtration?

Home Assignment

Ques: How much solid solute can be dissolved in certain fixed quantity of a solvent?

Signature of the

Assessment of Lesson

Name of Intern: Julie Koshy Date 24/1/2020
 Topic: Separating Insoluble Solids from Liquids Lesson Plan No. 11

Category Under Supervision		Rating Scale				
		Not Used	Poor	Avg.	Good	Very Good
Lesson Plan	1. Objectives stated in behavioural terms	0	1	2	3	4
Introduction	2. Student's previous knowledge tested	0	1	2	3	4
	3. Topic was announced effectively	0	1	2	3	4
Presentation	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
	9. Skill of Illustration with examples	0	1	2	3	4
Use of Teaching Aid	10. Adequate & appropriate	0	1	2	3	4
	11. Quality of teaching aid	0	1	2	3	4
Chalk Board Work	12. Organized and adequate	0	1	2	3	4
	13. Appropriateness and clarity	0	1	2	3	4
Management of the Class	14. Class environment & discipline	0	1	2	3	4
	15. Supervision of student's work	0	1	2	3	4
Evaluation	16. Skill of Closure	0	1	2	3	4
	17. Based on the behavioral objectives	0	1	2	3	4
	18. Quality of home assignment	0	1	2	3	4

For all comments and suggestions:-

(Use Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of lesson with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Lesson Plan No. 12

Date 25.01.2020

Name of the Internship School National Victor Public School

Class & Section VI-C Period 4th Duration 40 mins

Subject Science Sub-Subject _____

Topic The Living Organisms and their Surroundings

Specific Objectives after my class, students will be able to

- define a habitat
- describe the different types of habitat
- differentiate between Biotic & Abiotic components of a habitat.

Previous Knowledge

Students have general idea about their surroundings and the types of living organisms that live in a particular surrounding.

Introduction

Ques : What do you mean by surrounding?

Ans : Environment of a place in which an organism lives.

Ques : What is a habitat?

Ans : A place where an organism lives.

Ques : List the components of a habitat.

Ans : No Response.

Announcement of the Topic

Students, today we'll discuss about "The Living Organisms and their surroundings".

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Organisms and their surroundings	Student explains why different creatures live in different places.	<p><u>Teacher's Statement</u></p> <p>'Surrounding' means the environment of a place in which an organism lives. Different locations have different surroundings. The kind of organisms that live in a particular place depends on the surroundings of that place. This is why we see different creatures in different places.</p> <p>Eg: Camels are found in deserts while fishes & crabs are found in the sea.</p>	Student listens carefully and notes down in their notebook.	<p>Lecture Method Boardwork</p> <p>Class: VI Subject: Science Topic: THE LIVING ORGANISMS AND THEIR SURROUNDINGS</p> <p>The kind of organisms that live in a particular place depends on the surroundings of that place. Eg: Camels are found in deserts, fishes - sea.</p>	Explain why different creatures live in different places.
2. Habitat	Student defines the term habitat.	<p><u>Teacher's Statement</u></p> <p>All living organisms need a place to live in addition to food, water and air.</p> <p>A place where an organism lives is called its habitat. A habitat must provide the following:-</p> <ul style="list-style-type: none"> - food - shelter and - favourable climatic conditions to survive, breed & flourish. <p>The two components of a habitat are:</p> <ol style="list-style-type: none"> Biotic Components - Animals Abiotic Components - Non-living things eg: soil, water bodies, sunlight, rainfall etc. 	Student listens carefully and notes down in their notebook.	<p>Lecture Method Boardwork</p> <p>Class: VI Subject: Science Topic: THE LIVING ORGANISMS AND THEIR SURROUNDINGS</p> <p>HABITAT - A place where an organism lives</p> <p>↓ 2 COMPONENTS</p> <p>BIOTIC eg: Animals, plants etc</p> <p>ABIOTIC eg: soil, water bodies etc</p>	Define the term habitat. Give examples of Biotic & Abiotic components.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
3. Types of Habitat	Student describe the different types of habitat.	<u>Teacher's Statement</u> There are 3 main habitats in the biosphere.	Student listens carefully and writes down the different types of habitat with their examples in their notebook	Lecture Method Question Answer Method Board work	
• Terrestrial Habitat		i) <u>Terrestrial Habitat</u> (land) The organisms that live, grow and propagate on land are called terrestrial organisms. Their habitat is called Terrestrial habitat or land.	eg: Forest, Mountain, Desert - Terrestrial Habitat.	Class: VI Subject: Science Topic: THE LIVING ORGANISMS AND THEIR SURROUNDINGS 3 TYPES OF HABITAT	
• Aquatic Habitat		Can you give any examples of terrestrial habitat? ii) <u>Aquatic Habitat</u> The organisms that live and grow in water are called aquatic organisms and their habitat is called Aquatic Habitat.	Aquatic Organisms - Fish, Whales etc		
• Aerial or Arboreal Habitat		Give few examples of Aquatic organisms & aquatic habitat. iii) <u>Aerial or Arboreal habitat</u> The organisms that use trees for their activities are known as Arboreal organisms. eg: All birds - Eagle, Swallow	Aquatic Habitat - Ocean, Seas, Rivers etc		
		Organisms that use both water & land as their habitat are known as Amphibian Organisms. eg: Frog, Crocodile, Bufo etc.			Describe the different types of habitat.

Closure

To students, today we discussed about

- Habitat
- Components of Habitat
- Types of Habitat

Ques: Name the place where an organism lives to survive, reproduce & flourish.

Ques: Name the organisms that use trees for their activities.

Ques: Name the biotic components of a habitat

Ques: Name 2 organisms belonging to aquatic habitat.

Home Assignment

Ques: Distinguish between aquatic & terrestrial habitats.

Assessment of Lesson

Name of Intern: Julie Kostly Date: 25/1/2020
 Topic: The living organisms and their surroundings Lesson Plan No. 12

Category Under Supervision	Rating Scale					
	Not Used	Poor	Avg	Good	Very Good	
Lesson Plan	0	1	2	3	4	
Introduction	1. Objectives stated in behavioural terms	0	1	2	3	4
	2. Student's previous knowledge tested	0	1	2	3	4
	3. Topic was announced effectively	0	1	2	3	4
Presentation	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
	9. Skill of Illustration with examples	0	1	2	3	4
	10. Adequate & appropriate	0	1	2	3	4
	11. Quality of teaching aid	0	1	2	3	4
	12. Organized and adequate	0	1	2	3	4
Use of Teaching Aid	0	1	2	3	4	
Chalk Board Work	13. Appropriateness and clarity	0	1	2	3	4
	14. Class environment & discipline	0	1	2	3	4
Management of the Class	15. Supervision of student's work	0	1	2	3	4
	16. Skill of Closure	0	1	2	3	4
Evaluation	17. Based on the behavioural objectives	0	1	2	3	4
	18. Quality of home assignment	0	1	2	3	4

Overall comments and suggestions:-
 (Please Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of lesson with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Signature of the

Lesson Plan No. 13

Date 21.01.2020

Name of the Internship School National Victor Public School

Class & Section VI-D Period 2nd Duration 40 mins

Subject Science Sub-Subject Physics

Topic Motion and its Types

Specific Objectives After my class, students will be able to

- define the terms rest & motion.
- state the different types of motion.
- describe the different types of motion.

Previous Knowledge

Students have general idea about living and non-living things around.

Introduction

Q: Name some objects you see around.

A: A table, clock, tree, house, butterfly, a dog etc.

Q: Which of these are at rest?

A: House, table, clock.

Q: Which of these are moving?

A: A flying butterfly, a dog.

Q: Objects which are not in rest are said to be in _____.

A: No Response.

Announcement of the Topic

"Today, we'll discuss about 'Motion and its Types'."

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Motion	Student defines motion.	Teacher's Statement	Student listens carefully and writes down the important points in their notebook.	Feature Method	
• Definition	Student differentiates between motion & rest.	The objects which change their positions with time with respect to their surrounding are said to be in motion.		Question-Answer Method	
• Examples		An object is said to be at rest there is no change in its position with respect to its surrounding a given time.		Boardwork	
• Activity		eg: Ball dropping from a height, soldiers marching in parade, movement of bicycle on road etc. are said to be in motion.		Class: VI Subject: Science Topic: MOTION AND ITS TYPES	
		Pupil teacher divides the class into appropriate groups and each group is given a ball, chart & markers.		MOTION - Objects which change their positions with time with respect to their surroundings.	
		Pupil teacher instructs and asks questions in between:-	Student follows the instruction	REST - Objects which don't change their positions with time.	
		i) Place the ball and the car on the table.	Places the ball & car on the table.	Therefore, a body may be at rest with respect to one set of surroundings and at the same time be in motion with respect to another set of surroundings.	
		ii) Is the ball changing its position with respect to the car?	No		
		iii) Now move the ball.	Moves the ball, keeping the car fixed.		
		iv) If the ball is changing its position, what is the state of ball with respect to car?	The ball is said to be in motion.		
		v) What is the state of car with respect to ball?	The car is said to be at rest.		
		Therefore, a body may be at rest with respect to one set of surroundings and at the same time be in motion with respect to another set of surroundings.			Define motion. Differentiate between motion & rest.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities
2. Types of Motion	Student describes the different types of motion	Teacher's Statement Are all motions alike? No There are different kinds of motion classified as follows:- i) Translatory Motion: The motion in which a body moves as a whole and every part of it moves in the same direction. It is 2 types:- a) Rectilinear Motion: Movement of a body along a straight line. eg: An apple falling from a tree, soldiers marching in a parade. b) Curvilinear Motion: Movement of a body along a curved path. eg: Running race along a track, bus moving along a curved path etc. ii) Rotatory Motion: The motion of a body in which every part of it moves along a circular path about a fixed axis. eg: Motion of a spinning top, Motion of blades of a fan. iii) Oscillatory Motion: The to-and-fro motion of a body along the same path. eg: Motion of pendulum of a wall clock, A child on a swing.
- Translatory Motion		
• Rectilinearity		
• Curvilinearity		
- Rotatory Motion		
- Oscillatory Motion		

Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
Student listens carefully and notes down the different types of motion along with some examples in their notebook.	Lecture Method Deductive Method Boardwork Class: VI Subject: Science Topic: MOTION AND ITS TYPES Motions are classified as follows:- i) TRANSLATORY MOTION RECTILINEAR MOTION CURVILINEAR MOTION eg: Apple falling from a tree eg: bus moving along a curved path ii) ROTATORY MOTION eg: Motion of a spinning top iii) OSCILLATORY MOTION to & fro motion of a body eg: Motion of pendulum of a wall clock	Describe the different types of motion.

Closure

So students, today we discussed about
 - when is a body said to be in rest & when in motion.

- different types of motion.

Ques: Differentiate between Rectilinear & Curvilinear motion.

Ques: Give an example of Translatory Motion.

Ques: Define an oscillatory motion with an example.

Ques: Describe Rotational Motion.

Ques: What kind of motion is shown by the earth moving about its own axis?

Home Assignment

Ques: Are all the passengers sitting in a compartment of a running train at rest or in motion with respect to each other? Explain.

Signature of the Mentor

Assessment of Lesson

Name of Intern: Julie Koshy
 Topic: Motion and its Types
 Date: 27/1/2020
 Lesson Plan No. 13

Lesson Plan	Category Under Supervision	Rating Scale				
		Not Used	Poor	Avg.	Good	Very Good
Introduction	1. Objectives stated in behavioural terms	0	1	2	3	4
	2. Student's previous knowledge tested	0	1	2	3	4
	3. Topic was announced effectively	0	1	2	3	4
Presentation	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
	9. Skill of Illustration with examples	0	1	2	3	4
	10. Adequate & appropriate	0	1	2	3	4
Use of Teaching Aid	11. Quality of teaching aid	0	1	2	3	4
	12. Organized and adequate	0	1	2	3	4
Chalk Board Work	13. Appropriateness and clarity	0	1	2	3	4
	14. Class environment & discipline	0	1	2	3	4
Management of the Class	15. Supervision of student's work	0	1	2	3	4
	16. Skill of Closure	0	1	2	3	4
Evaluation	17. Based on the behavioural objectives	0	1	2	3	4
	18. Quality of home assignment	0	1	2	3	4

Over all comments and suggestions:-
 Dress Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of lesson with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Galgotias University Uttar Pradesh, G.B. Nagar
School of Education

Lesson Plan No. 14

Date 28.01.2020

Name of the Internship School National Victor Public School

Class & Section VI-C Period 4th Duration 40 mins

Subject Science Sub-Subject _____

Topic Adaptation in Animal & Plants

Specific Objectives After my class, students will be able to

- Define adaptation.
- Differentiate between permanent & temporary adaptation.
- Explain adaptation to Desert habitat.

Previous Knowledge

Students have general idea about habitat, types and components of habitat.

Introduction

Ques: Name the place where an organism lives to survive, reproduce & flourish.

Ans: Habitat

Ques: What are the three types of Habitat?

Ans: Terrestrial Habitat, Aquatic Habitat & Aerial Habitat

Ques: Give 3 examples of Terrestrial Habitat.

Ans: Grasslands, Deserts, Mountains etc.

Ques: _____ is referred to as the tendency of an organism to bring about changes in the structure & function of its body to suite to the given environment.

Ans: No Response.

Announcement of the Topic

Today we will discuss about 'Adaptation in Animals and Plants'.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Adaptation	Student defines adaptation with examples	<p><u>Teacher's Statement</u></p> <p>The tendency of an organism to develop certain specific features which improve the chances of its survival in the environment in which it lives is known as adaptation.</p> <p>In response to a particular environment, an organism brings about certain changes in the structure & function of its body and makes it well suited to the environment. For example: Camel & Desert rat are adapted to the desert habitat.</p>	Student listens carefully and notes down in their notebook.	<p>Lecture Method</p> <p>Boardwork</p> <p>Class: VI Subject: Science</p> <p>Topic: ADAPTATION IN PLANTS AND ANIMALS</p> <p>Adaptation - Tendency of an organism to develop specific features which improve its chances of survival in the env. in which it lives.</p>	Define adaptation with examples.
• Meaning & Definition					
• Examples					
2. Types of Adaptation	Student differentiates between Permanent adaptation and Temporary adaptation.	<p><u>Teacher's Statement</u></p> <p>Adaptation is of two types -</p> <p>i) Permanent Adaptation :- It has a genetic basis and causes permanent changes in the individuals. eg: Camel is well adapted to desert environment and not to water. A fish is adapted to live in water.</p> <p>ii) Temporary Adaptation :- It is a short term adaptation and is not inherited. eg: A bird adapted to live in the cage is a temporary adaptation.</p>	Student listens carefully and notes the 2 types of adaptation in their notebook.	<p>Lecture Method</p> <p>Deductive Method</p> <p>Boardwork</p> <p>Class: VI Subject: Science</p> <p>Topic: ADAPTATION IN PLANTS & ANIMALS</p> <pre> graph TD A[ADAPTATION] --> B[PERMANENT ADAPTATION] A --> C[TEMPORARY ADAPTATION] </pre>	Diff. between Permanent and Temporary Adaptation.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
3. Adaptation to Desert Habitat	Student explains adaptation to Desert habitat	<p>Pupil Teacher asks the following questions:-</p> <p>Q. Define a terrestrial habitat.</p> <p>Q. Give few examples of terrestrial habitat.</p> <p>Some terrestrial habitats are grassland, forests, deserts, mountains etc. let us discuss a few of these.</p> <p><u>Teacher's Statement</u></p> <p><u>Adaptation to Desert Habitat</u></p> <p><u>Desert Animals:-</u> The climate of deserts is dry and hot. Deserts have very little vegetation.</p> <ul style="list-style-type: none"> Many animals living in deserts remain inactive during the day and become active during the night. This prevents loss of water from the body. Animals that show this type of adaptation are called Nocturnal. Certain animals hide themselves in burrows with gelatinous secretions and sleep for eight to nine months in a year. This is called hibernation. In summers, the cold-blooded animals like lizards, snakes & Mosquitoes are more active during night. In winters, they enjoy sun during the day and when cold they become inactive for a certain long period. 	<p>The plants and animals that live on land are said to live in terrestrial habitat.</p> <p>eg: Deserts, Mountains etc</p> <p>Student listens carefully and notes down in their notebook.</p>	<p>Question-Answer Method.</p> <p>Recurrence Method.</p> <p>Boardwork</p> <p>Class: VI Subject: Science</p> <p>Topic: ADAPTATION IN PLANTS & ANIMALS</p> <p><u>Terrestrial Habitat</u> eg: grassland, forests, deserts, mountains etc.</p> <p><u>ADAPTATION TO DESERT HABITAT</u></p> <p>NOCTURNAL ANIMALS - INACTIVE during the day and ACTIVE in the night</p> <p>HIBERNATION - Animals hide themselves in burrows with gelatinous secretions and sleep for eight to nine months in a year</p>	<p>Explain adaptation to Desert Habitat</p>

Closure

So students, today we discussed about
 - adaptation in animals & plants
 - 2 types of adaptation: Permanent & Temporary Adaptation
 Ques: Differentiate between Permanent & Temporary Adaptation
 Ques: Define hibernation.
 Ques: Why do most animals living in deserts remain inactive during the day & active during the night?

Home Assignment

Ques: How do desert animals adapt themselves to the conditions in the desert?

Signature of the Mentor

Assessment of Lesson

Name of Intern: Julie Kochy Date: 28.01.2020
 Topic: Adaptation in Plants and Animals Lesson Plan No. 14

Category Under Supervision		Rating Scale				
		Not Used	Poor	Avg.	Good	Very Good
Lesson Plan	1. Objectives stated in behavioural terms	0	1	2	3	4
	2. Student's previous knowledge tested	0	1	2	3	4
	3. Topic was announced effectively	0	1	2	3	4
Introduction	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
	9. Skill of Illustration with examples	0	1	2	3	4
Use of Teaching Aid	10. Adequate & appropriate	0	1	2	3	4
	11. Quality of teaching aid	0	1	2	3	4
Chalk Board Work	12. Organized and adequate	0	1	2	3	4
	13. Appropriateness and clarity	0	1	2	3	4
Management of the Class	14. Class environment & discipline	0	1	2	3	4
	15. Supervision of student's work	0	1	2	3	4
Evaluation	16. Skill of Closure	0	1	2	3	4
	17. Based on the behavioral objectives	0	1	2	3	4
	18. Quality of home assignment	0	1	2	3	4

For all comments and suggestions:-
 Dress Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of lesson with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Galgotias University Uttar Pradesh, G.B. Nagar
School of Education

Lesson Plan No. 15.

Date 29.01.2020

Name of the Internship School National Victor Public School

Class & Section VI-C Period 4th Duration 40 mins

Subject Science Sub-Subject _____

Topic Adaptation in Animals & Plants.

Specific Objectives After my class, students will be able to

- write two characteristics of adaptation of camels in deserts
- explain the adaptation in Desert plants
- describe adaptation in Mountainous regions.

Previous Knowledge

Students have general idea about habitat, types of habitat and adaptation.

Introduction

Ques : What do you mean by adaptation?

Ans : The tendency of an organism to develop certain specific features which improve the chances of its survival in the given environment.

Ques : Give an example of Permanent Adaptation.

Ans : Camel is well adapted to desert-environment.
A fish is well adapted to live in water.

Ques : How is the camel well adapted to the desert conditions?

Ans : No Response.

Announcement of the Topic

Students, today we'll discuss about 'Adaptation in Animals & Plants'.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Adaptation in Camel.	Student identifies the characteristics the camel has adapted to survive in deserts.	<p><u>Teacher's Statement</u></p> <p>Camel is the only large animal that survives in desert conditions. Camel is called 'ship of the desert'.</p> <p>It is adapted to the desert conditions because of the following :-</p> <ol style="list-style-type: none"> i) Its humps contain large pads which helps it to move on hot & slippery sand. ii) The hump of the camel is a reservoir of food in the form of fat. iii) It drinks a large quantity of water and stores it in its cells, muscles and connective tissues. iv) It can live without drinking water for about 2 weeks. It excretes very little water in the form of urine. v) It can adjust its internal temperature according to the surrounding air. It sweats profusely in the day sun to maintain its body temperature. 	Student listens carefully and notes down the characteristics the camel has adapted to survive in deserts in their notebook.	<p>Lecture Method</p> <p>Boardwork</p> <p>Class: VI Subject: Science</p> <p>Topic: ADAPTATION IN PLANTS AND ANIMALS</p> <p><u>Adaptation in Camel</u></p> <ul style="list-style-type: none"> → Humps contain large pads - helps to move in sand. → Hump of the camel - reservoir of food. → Drinks a large quantity of water & stores it in cells, muscles & connective tissues. → Can adjust its internal temperature according to the surrounding air. 	Identify the characteristics the camel has adapted to survive in the deserts.
2. Desert Plants	Student explains the adaptation in desert plants.	<p><u>Teacher's Statement</u></p> <p>The plants show the following characteristics to survive in hot & dry climate.</p> <ol style="list-style-type: none"> i) Their leaves are reduced to scales or spines which reduce the loss of water by transpiration. 	Student listens carefully and notes in their notebook.	Lecture Method	

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
		<p>ii) The leaves bear sunken stomata to reduce the rate of transpiration.</p> <p>iii) The thick cuticle is found on the stem and leaves for the prevention of loss of water.</p> <p>The plants that grow under very poor water conditions, such as in deserts are called xerophytes.</p> <p>eg: Cactus & Opuntia.</p>			<p>Explain the adaptation in desert plants.</p>
3. Adaptation to Mountainous Regions	Student describes the adaptation by Mountain Plants and Mountain Animals.	<p>Teacher's Statement</p> <p>Mountainous regions are generally very cold and windy.</p> <p>Mountain Plants/Trees:-</p> <p>The trees adapted to such cold snowy climate</p> <ul style="list-style-type: none"> - are cone-shaped & have sloping branches. - have needle-like leaves <p>These adaptations help the rain and snow to slide off easily.</p> <p>Mountain Animals :-</p> <p>The animals in these regions are usually white or light colour. This helps them to camouflage with the surroundings. This also helps them in thermal regulation.</p> <p>They have thick skin/fur to protect them from cold.</p>	<p>Student listens carefully and writes down in their notebook.</p>	<p>Lecture Method</p> <p>Boardwork</p> <p>Class: VI Subject: Science</p> <p>Topic: ADAPTATION IN PLANTS AND ANIMALS</p> <p>MOUNTAIN PLANTS/TREES</p> <ul style="list-style-type: none"> - cone-shaped - sloping branches - needle-like leaves <p>MOUNTAIN ANIMALS</p> <ul style="list-style-type: none"> - white/light coloured - thick skin/fur, helps in thermal regulation 	<p>Describe the adaptation by Mountain plants & Mountain Animals.</p>

Closure

To students, today we learnt about

- adaptation in camel
- adaptation in desert plants
- adaptation to mountainous regions.

Ques: Write two characteristics which the camel has adapted to survive in deserts.

Ques: What are the common features found in mountain animals?

Home Assignment

Ques: How do desert plants adapt themselves to the conditions in the desert?

Assessment of Lesson

Name of Intern: Julie Koshy Date: 29.01.2020
 Topic: Adaptation in Animals & Plants Lesson Plan No. 15

Category Under Supervision		Rating Scale				
		Not Used	Poor	Avg.	Good	Very Good
Lesson Plan	1. Objectives stated in behavioural terms	0	1	2	3	4
	2. Student's previous knowledge tested	0	1	2	3	4
	3. Topic was announced effectively	0	1	2	3	4
Introduction	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
	9. Skill of Illustration with examples	0	1	2	3	4
Presentation	10. Adequate & appropriate	0	1	2	3	4
	11. Quality of teaching aid	0	1	2	3	4
Use of Teaching Aid	12. Organized and adequate	0	1	2	3	4
	13. Appropriateness and clarity	0	1	2	3	4
Chalk Board Work	14. Class environment & discipline	0	1	2	3	4
	15. Supervision of student's work	0	1	2	3	4
Management of the Class	16. Skill of Closure	0	1	2	3	4
	17. Based on the behavioral objectives	0	1	2	3	4
Evaluation	18. Quality of home assignment	0	1	2	3	4

For all comments and suggestions:-
 Lesson Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of lesson with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Signature of the Mentor

Galgotias University Uttar Pradesh, G.B. Nagar
School of Education

Lesson Plan No. 16

Date 30.01.2020

Name of the Internship School National Victor Public School

Class & Section X-D Period 2nd Duration 40 mins

Subject Science Sub-Subject Chemistry

Topic Chemical Equations

Specific Objectives

After my class, students will be able to

- define a chemical equation.
- balance a chemical equation.

Previous Knowledge

Students have some knowledge about chemical reactions.

Introduction

Q: What happens when wood is burned?

Ans: It changes into ashes.

Q: What reaction wood or coal undergoes when it is burned?

Ans: Chemical Reaction

Q: How can we denote a chemical reaction in symbolic form?

Ans: No Response.

Announcement of the Topic

Today we will discuss about "Chemical equation".

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Chemical equation - Meaning - Definition	Student defines a chemical equation.	<p><u>Teacher's Statement</u></p> <p>A chemical equation consists of chemical formulas of the reactants and the chemical formula of the products.</p> <p>The two are separated by an arrow symbol and each individual substance's chemical formula is separated from others by a plus sign.</p>	Student listens carefully.	<p>Boardwork</p> <p>Class: X Subject: Science</p> <p>Topic: CHEMICAL EQUATIONS</p> <p>Chemical equation of Hydrochloric acid with Sodium</p> $2\text{HCl} + 2\text{Na} \rightarrow 2\text{NaCl} + \text{H}_2$ <p>REACTANTS PRODUCTS</p>	
2. Common Symbols	Student differentiates between different types of reactions.	<p><u>Teacher's Statement</u></p> <p>Symbols are used to differentiate between different types of reactions.</p> <p>"\rightarrow" symbol is used to denote a net forward reaction.</p> <p>"\rightleftharpoons" symbol is used to denote a reaction in both directions.</p> <p>"\rightleftharpoons" symbol is used to denote equilibrium.</p> <p>when stating physical state, (s) - denotes a solid (l) - denotes a liquid (g) - denotes a gas (aq) - denotes an aqueous solution</p> <p>Δ (delta) - energy in the form of heat is added to the reaction. It is indicated above an arrow.</p>	Student listens carefully and writes down in their notebook.	<p>Lecture Method</p> <p>Boardwork</p> <p>Class: X Subject: Science</p> <p>Topic: CHEMICAL EQUATIONS</p> <p>"\rightarrow" Forward Reaction "\rightleftharpoons" equilibrium (s) - solid state (l) - liquid state (g) - gaseous state (aq) - aqueous state</p>	Define a Chemical equation.
					Differentiate between " \rightarrow ", " \rightleftharpoons ", (s), (l), and (g) symbols used in a chemical equation.

Presentation

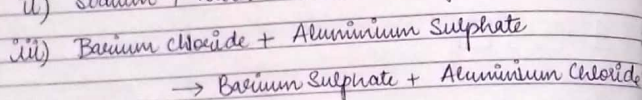
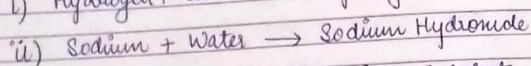
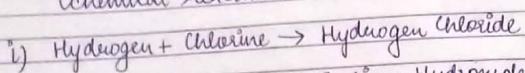
Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation												
3. Balancing a Chemical equation	Student balances a chemical equation using the steps	<p>Statement:</p> <p>A chemical equation is balanced so that the numbers of atoms of each type involved in a chemical reaction are the same on the reactants and products sides of the equation.</p> <p>Steps to balance a chemical equation :-</p> <p>Step 1: Write the chemical equation of Iron reacting with water and draw boxes around the formula.</p> <p>Step 2: List the number of atoms of different elements present in unbalanced equation.</p> <p>Step 3: Start balancing with the compound that contains maximum number of atoms.</p> <p>Step 4: Balance Fe and H atoms.</p> <p>Step 5: Count atoms of each element on both sides of the equation.</p> <p>Step 6: Writing symbols of Physical state.</p>	<p>Student listens carefully and follows the steps provided.</p> <p> $\text{Fe} + \text{H}_2\text{O} \rightarrow \text{Fe}_3\text{O}_4 + \text{H}_2$ <table border="1"> <thead> <tr> <th>Element</th> <th>No. of atoms (LHS)</th> <th>No. of atoms (RHS)</th> </tr> </thead> <tbody> <tr> <td>Fe</td> <td>1</td> <td>3</td> </tr> <tr> <td>H</td> <td>2</td> <td>2</td> </tr> <tr> <td>O</td> <td>1</td> <td>4</td> </tr> </tbody> </table> $3\text{Fe} + 4\text{H}_2\text{O} \rightarrow \text{Fe}_3\text{O}_4 + 4\text{H}_2$ </p>	Element	No. of atoms (LHS)	No. of atoms (RHS)	Fe	1	3	H	2	2	O	1	4	<p>Question-Answer Method</p> <p>Boardwork</p> <p>Class: X Subject: Science Topic: CHEMICAL EQUATIONS</p> <p> $\text{Fe} + \text{H}_2\text{O} \rightarrow \text{Fe}_3\text{O}_4 + \text{H}_2$ $\text{Fe} + 4\text{H}_2\text{O} \rightarrow \text{Fe}_3\text{O}_4 + \text{H}_2$ $\text{Fe} + 4\text{H}_2\text{O} \rightarrow \text{Fe}_3\text{O}_4 + 4\text{H}_2$ $3\text{Fe} + 4\text{H}_2\text{O} \rightarrow \text{Fe}_3\text{O}_4 + 4\text{H}_2$ </p> <p>∴ The given equation is balanced.</p>	<p>Balance the chemical equation</p> <p>$\text{Fe} + \text{H}_2\text{O} \rightarrow \text{Fe}_3\text{O}_4 + \text{H}_2$</p>
Element	No. of atoms (LHS)	No. of atoms (RHS)															
Fe	1	3															
H	2	2															
O	1	4															

Assessment of Lesson

Closure

So students, today we discussed about chemical equations and how to balance a given chemical equation.

Ques: Write the balanced equation for the following chemical reactions:-



Home Assignment

Ques: Write a balanced chemical equation for the given reaction:- Sodium hydroxide reacts with hydrochloric acid to produce sodium chloride solution & water.

Name of Intern: Julie Koshy Date 30-01-2020
 Topic: Chemical Equations Lesson Plan No. 16

Category Under Supervision		Rating Scale				
		Not Used	Poor	Avg.	Good	Very Good
Lesson Plan Introduction	1. Objectives stated in behavioural terms	0	1	2	3	4
	2. Student's previous knowledge tested	0	1	2	3	4
	3. Topic was announced effectively	0	1	2	3	4
Presentation	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
	9. Skill of Illustration with examples	0	1	2	3	4
Use of Teaching Aid	10. Adequate & appropriate	0	1	2	3	4
	11. Quality of teaching aid	0	1	2	3	4
Chalk Board Work	12. Organized and adequate	0	1	2	3	4
	13. Appropriateness and clarity	0	1	2	3	4
Management of the Class	14. Class environment & discipline	0	1	2	3	4
	15. Supervision of student's work	0	1	2	3	4
Evaluation	16. Skill of Closure	0	1	2	3	4
	17. Based on the behavioral objectives	0	1	2	3	4
	18. Quality of home assignment	0	1	2	3	4

For all comments and suggestions:-

Classroom Management, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of lesson with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Signature of the M

Lesson Plan No. 17

Date 31-01-2020

Name of the Internship School National Victor Public School

Class & Section X-A Period _____ Duration _____

Subject Science Sub-Subject Chemistry

Topic Types of Chemical Reactions

Specific Objectives After my class, students will be able to

- Define a combination reaction.
- Differentiate between a combination & decomposition reaction
- Explain double combination reaction

Previous Knowledge

Students have knowledge about chemical reactions and how to balance a given chemical equation.

Introduction

Ques : What happens when coal is burned in air?

Ans : It forms carbon dioxide.

Ques : Write the chemical equation involved.

Ans : $C + O_2 \rightarrow CO_2$

Ques : What type of reaction is this?

Ans : No Response.

Announcement of the Topic

Students, today we'll discuss about "Types of Chemical Reactions".

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Combination Reaction	Student defines a combination reaction.	<u>Teacher's Statement</u> Combination reaction is a reaction in which two or more substances react together to form just a product. eg: Burning of Coal $C(s) + O_2(g) \rightarrow CO_2(g)$ Formation of water $2H_2(g) + O_2(g) \rightarrow 2H_2O(l)$	Student listens carefully and notes down in their notebook.	Lecture Method	
- Definition					Define a combination reaction.
- Example					
2. Decomposition Reaction	Student defines decomposition reaction.	<u>Teacher's Statement</u> Decomposition Reaction is a reaction in which one reactant breaks up to form two or more simpler products. eg: Decomposition of Ferric sulphate gives ferric oxide, sulphur dioxide and sulphur trioxide $2FeSO_4 \rightarrow Fe_2O_3 + SO_2 + SO_3$	Student listens carefully and notes down in their notebook.	Lecture Method Boardwork Class: X Subject: Science Topic: Types of Chemical Reactions DECOMPOSITION REACTION - one reactant breaks up to form two or more simpler products $2FeSO_4 \xrightarrow{\text{Heat}} Fe_2O_3 + SO_2 + SO_3$	Define a decomposition reaction.
- Definition					
- Example					

Presentation

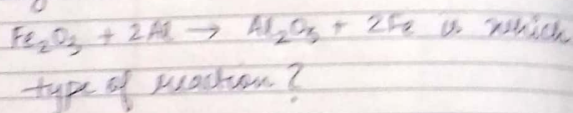
Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
5. Displacement Reaction	Student describes displacement reaction with examples	<p><u>Teacher's Statement</u></p> <p>A reaction in which a more reactive element displaces a less reactive one from its compound is called a displacement reaction.</p> <p>Example :- When magnesium placed in hydrochloric acid, magnesium replaces hydrogen.</p> <p>This is because magnesium is more reactive than hydrogen.</p> <p>Pupil Teacher asks the following questions :-</p> <p>i) Write a balanced chemical equation when magnesium reacts with hydrogen.</p> <p>ii) Write a balanced chemical equation when iron reacts with copper sulphate solution.</p> <p>iii) In the above reactions, which element replaces which element?</p> <p>iv) Why does the blue colour of CuSO_4 solution fade when iron nail is placed in it?</p>	<p>Student listens carefully and writes it in their notebook.</p> <p>$\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$</p> <p>$\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$</p> <p>Fe</p> <p>Blue colour of CuSO_4 fades when iron is placed in its solution because Fe is more reactive than copper.</p>	<p>Question-Answer Method</p> <p>Boardwork</p> <p>Class: X Subject: Science</p> <p>Topic: Types of Chemical Reactions</p> <p>DISPLACEMENT REACTION</p> <p>$\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$</p> <p>$\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$</p>	<p>Write a balanced chemical equation when zinc reacts with copper sulphate solution.</p>

To students, today we discussed about different types of chemical reactions.

i) Why does the colour of copper sulphate solution change when an iron nail is dipped in it?

ii) Give an example of double displacement reaction.

iii) The given reaction



Home Assignment

Ques: Translate the following statement into chemical equation & balance it :-

Potassium metal reacts with water to give potassium hydroxide & hydrogen gas.

Date 31.01.2020
 of Intern Julie Koshy
 Types of Chemical Reactions Lesson Plan No. 17

Category Under Supervision	Rating Scale					
	Not Used	Poor	Avg.	Good	Very Good	
Lesson Plan Introduction	1. Objectives stated in behavioural terms	0	1	2	3	4
	2. Student's previous knowledge tested	0	1	2	3	4
	3. Topic was announced effectively	0	1	2	3	4
Presentation	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
	9. Skill of Illustration with examples	0	1	2	3	4
Use of Teaching Aid	10. Adequate & appropriate	0	1	2	3	4
	11. Quality of teaching aid	0	1	2	3	4
Chalk Board Work	12. Organized and adequate	0	1	2	3	4
	13. Appropriateness and clarity	0	1	2	3	4
Management of the Class	14. Class environment & discipline	0	1	2	3	4
	15. Supervision of student's work	0	1	2	3	4
Evaluation	16. Skill of Closure	0	1	2	3	4
	17. Based on the behavioral objectives	0	1	2	3	4
	18. Quality of home assignment	0	1	2	3	4

all comments and suggestions:-

Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of (with the help of Students)

Signature of the Intern

Signature of the Mentor

Signature of the M

Lesson Plan No. 18

Date 3.02.2020

Name of the Internship School National Victor Public School

Class & Section X-C Period 3rd Duration 40 mins

Subject Science Sub-Subject Chemistry

Topic Oxidation and Reduction Reactions

Specific Objectives After my class, students will be able to

- define oxidation & reduction reaction.
- explain oxidation and reduction reaction with examples.
- describe the effects of oxidation and reduction in daily life.

Previous Knowledge

Students have knowledge about different types of chemical reactions.

Introduction

Ques : When zinc reacts with oxygen, what product is formed?

Ans : Zinc Oxide (ZnO)

Ques : Write a reaction for copper oxide reacting with hydrogen gas.

Ans : $\text{CuO} + \text{H}_2 \rightarrow \text{Cu} + \text{H}_2\text{O}$

Ques : In this reaction, CuO gets reduced to Cu & H₂ gets oxidised to H₂O so what is such reaction known as?

Ans : No Response

Announcement of the Topic

Today, we'll discuss about "Oxidation and Reduction Reactions".

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
3. Redox Reaction	Student defines redox reaction	Teacher's Statement			
- Definition		Reactions in which oxidation and reduction occur simultaneously are called Redox Reactions	Student listens carefully and notes down in their notebook.	Lecture Method	
- Oxidising & Reducing Agent	Student differentiates between oxidising and reducing agent	<p>Oxidising Agent :- The substance which give oxygen or substance which removes hydrogen.</p> <p>Reducing Agent :- The substance which give hydrogen or remove oxygen</p> <p>Pupil teacher asks the following questions:-</p> <p>i) Write the reaction between $\text{CuO} + \text{H}_2 \rightarrow \text{Cu} + \text{H}_2\text{O}$ Copper Oxide & Hydrogen (gas)</p> <p>ii) In the above reaction, which substance undergoes oxidation?</p> <p>iii) Which substance undergoes reduction?</p> <p>iv) Which is the oxidising Agent?</p> <p>v) Which is the Reducing Agent?</p>		<p>Question-Answer Method</p> <p>Boardwork</p> <p>Class: X Subject: Science</p> <p>Topic: OXIDATION AND REDUCTION REACTIONS</p> <p> $\begin{array}{c} \text{Oxidation} \downarrow \\ \text{CuO} + \text{H}_2 \rightarrow \text{Cu} + \text{H}_2\text{O} \\ \uparrow \text{Reduction} \\ \text{(Oxidising Agent)} \end{array}$ </p>	
					Define Redox Reaction
					Name the oxidising & reducing agent in the given reaction $\text{H}_2\text{S} + \text{Cl}_2 \rightarrow \text{S} + 2\text{HCl}$

Closure

To students, today we discussed about

- Oxidation Reaction
- Reduction Reaction
- Redox Reaction
- Oxidising & Reducing Agent.

Ques: Write a balanced chemical equation for the chemical reaction between hydrogen sulphide and chlorine gas.

Name the oxidising & reducing agent in the above chemical equation.

Home Assignment

Ques: Find out atleast 5 impacts of oxidation and reduction reaction in everyday life

Assessment of Lesson

of Intern: Julie Koshy Date 3.02.2020
Oxidation & Reduction Reactions Lesson Plan No. 18

Category Under Supervision	Rating Scale				
	Not Used	Poor	Avg.	Good	Very Good
Lesson Plan	0	1	2	3	4
Introduction	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
Presentation	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
Use of Teaching Aid	0	1	2	3	4
	0	1	2	3	4
Chalk Board Work	0	1	2	3	4
	0	1	2	3	4
Management of the Class	0	1	2	3	4
	0	1	2	3	4
Evaluation	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4

all comments and suggestions:-
 Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of n with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Signature of the

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School of Education

Lesson Plan No. 19

Date 4.02.2020

Name of the Internship School National Victor Public School

Class & Section VI-C Period 4th Duration 40 mins

Subject Science Sub-Subject _____

Topic Adaptation to grasslands.

Specific Objectives After my class, students will be able to

- list the common features of grasslands.
- describe the adaptation to grassland in herbivores and in carnivores.

Previous Knowledge

Students have general idea about adaptation and its two types: Permanent- & Temporary Adaptation.

Introduction

Ques : what do you mean by terrestrial habitat?

Ans : Terrestrial habitat refers to land habitat.

Ques : Give few examples of terrestrial habitat.

Ans : Grasslands, Deserts, Mountains etc.

Ques : Name 2 animals found in grasslands.

Ans : Deer, horse, lion etc

Ques : How do these animals (deer, horse) save themselves from predators?

Ans : No Response.

Announcement of the Topic

Students, today we'll discuss about 'Adaptation to grasslands'.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Adaptation to grasslands	Student cites the features of grasslands.	<p><u>Teacher's Statement</u></p> <p>The grasslands cover nearly of the earth's land. Tropical grasslands are hot throughout the year, whereas the temperate grasslands are hot in summer and very cold in winters.</p> <p>All grasslands have the following common features :-</p> <ul style="list-style-type: none"> - Plenty of grass as food. - Continuity & openness - Lacking cover to hide from predators <p>eg: Animals : Deer, Horse, etc.</p>	Student listens carefully and notes down in their notebook.	<p>Lecture Method.</p> <p>Board work</p> <p>Class: VI Subject: Science</p> <p>Topic: ADAPTATION TO GRASSLANDS</p> <p>grasslands</p> <ul style="list-style-type: none"> ↳ Plenty of grass ↳ Continuity & openness ↳ Lacking cover to hide from predators 	Cite the feature of grasslands
2. Adaptation to grassland in Herbivores	Student describes the adaptation to grassland in Herbivores.	<p><u>Teacher's Statement</u></p> <p>Herbivores, such as deer, cow etc living in grasslands have the following adaptations :-</p> <ul style="list-style-type: none"> * Strong teeth for chewing & hard grass. * long ears to follow the movement of predator. A predator is an animal which makes them its prey. * long legs to run fast & save themselves from predators * eyes on the sides of their head which provide a wider field view to follow the movement of predators. 	Student listens carefully and notes down in their notebook.	<p>Lecture Method</p> <p>Board work</p> <p>Class: VI Subject: Science</p> <p>Topic: ADAPTATION IN GRASSLANDS</p> <p>Herbivores</p> <ul style="list-style-type: none"> - Strong teeth - long ears - long legs - eyes on sides of their head. 	Describe the adaptation to grassland in Herbivores.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
3. Adaptation to grassland in carnivores	Student explains the adaptation to grassland in carnivores	<p>Pupil Teacher asks the following questions:-</p> <p>Ques: Differentiate between Herbivores & Carnivores</p> <p>Ques: Give 2 examples of carnivores</p> <p><u>Teacher's Statement</u></p> <p>Carnivores like Lion, tiger etc are found in grasslands. They have the following adaptations:-</p> <ul style="list-style-type: none"> * Sharp & long claws in their legs to catch & hold and the body of their prey. * high-brown coloured body. This helps them to hide in grass. * the eyes in front of the face help them to hide in the grass and help in locating prey. * long & strong canines help them in tearing the body of their prey. 	<p>Herbivores</p> <ul style="list-style-type: none"> - Plant-eating - Animal-eating - lion, tiger <p>Student listens carefully and notes down in their notebook.</p>	<p>Question-Answer Method.</p> <p>Lecture Method.</p> <p>Boardwork</p> <p>Class: VI Subject: Science</p> <p>Topic: ADAPTATION TO GRASSLANDS</p> <p>Carnivores:-</p> <ul style="list-style-type: none"> ↳ Sharp & long claws. ↳ high-brown coloured body. ↳ eyes in front of the face ↳ long & strong canines 	<p>Explain the adaptation to grassland in carnivores</p>

Closure

So students, today we discussed about
 - Adaptation to grassland in Herbivores
 - Adaptation to grassland in Carnivores

Ques: Give one adaptation in herbivores living in grasslands

Ques: What are the common features of all grasslands?

Ques: Name an organ of the grassland herbivore which helps it to follow the movements of a predator.

Home Assignment

Ques: Describe the adaptation in herbivores which help them to save themselves from predators.

Signature of the Intern

Assessment of Lesson

Name of Intern: Julie Koshy Date: 4.02.2020
 Topic: Adaptation to grasslands Lesson Plan No. 19

Category Under Supervision		Rating Scale				
		Not Used	Poor	Avg.	Good	Very Good
Lesson Plan	1. Objectives stated in behavioural terms	0	1	2	3	4
Introduction	2. Student's previous knowledge tested	0	1	2	3	4
	3. Topic was announced effectively	0	1	2	3	4
Presentation	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
Use of Teaching Aid	9. Skill of Illustration with examples	0	1	2	3	4
	10. Adequate & appropriate	0	1	2	3	4
	11. Quality of teaching aid	0	1	2	3	4
Chalk Board Work	12. Organized and adequate	0	1	2	3	4
	13. Appropriateness and clarity	0	1	2	3	4
Management of the Class	14. Class environment & discipline	0	1	2	3	4
	15. Supervision of student's work	0	1	2	3	4
Evaluation	16. Skill of Closure	0	1	2	3	4
	17. Based on the behavioral objectives	0	1	2	3	4
	18. Quality of home assignment	0	1	2	3	4

All comments and suggestions:-
 is Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Lesson Plan No. 20

Date 5-02-2020

Name of the Internship School National Victor Public School

Class & Section VI-C Period 4th Duration 40 mins

Subject Science Sub-Subject _____

Topic Aquatic Adaptation in Animals & Plants

- Specific Objectives** After my class, students will be able to
- explain aquatic adaptation in animals and plants.
 - define hydrophytes.
 - describe the adaptation of hydrophytes.

Previous Knowledge

Students have general knowledge about adaptation and its types.

Introduction

Ques : What do you mean by aquatic habitat?

Ans : Habitat in which organisms live and breed in water is called aquatic habitat.

Ques : Give 4 examples of aquatic habitat.

Ans : Sea, lake, Ponds, Rivers etc.

Ques : Give 2 examples of aquatic animals.

Ans : Whales, Dolphins etc.

Ques : How do these animals adapt themselves to water?

Ans : No Response.

Announcement of the Topic

Students, today we'll discuss about 'Aquatic Adaptation in Animals & Plants'.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Aquatic Adaptation in Animals & Plants.	Student cites the adaptive features for aquatic adaptation.	<p>Pupil/Teacher asks the following questions:-</p> <p>Ques: Animals that live in water are said to live in _____ habitat.</p> <p>Ques: Give examples of aquatic habitat.</p> <p>Ques: Give examples of aquatic animals.</p> <p>Statement</p> <p>The adaptive features for aquatic adaptation are:-</p> <ul style="list-style-type: none"> - Body of aquatic animals is compressed laterally to reduce friction. This allows swift passage through water while swimming. - Fins in fish & flippers in whale help in swimming. - Frogs and duck have webbed feet for swimming. - Swim Bladder present in us filled with air and maintain buoyancy. - Gills are the organs of respiration. Gills exchange gases dissolved in water. - Dolphins and whales do not have gills. They breathe in air through nostrils or blowholes. 	<p>Students cite aquatic habitat</p> <p>e.g., lake, river, pond, sea, whales etc</p> <p>Student listens carefully and notes down in their notebook.</p>	<p>Question-Answer Method.</p> <p>Lecture Method</p> <p>Boardwork</p> <p>Class: VI Subject: Science</p> <p>Topic: AQUATIC ADAPTATION IN ANIMALS & PLANTS</p> <p>Adaptive features for aquatic adaptation:</p> <ul style="list-style-type: none"> → Body: Compressed laterally. → Fins in fish & flippers in whale - Swimming → Gills - Respiration → Swim bladder - Buoyancy 	<p>Evaluation</p> <p>Cite the adaptive features for aquatic adaptation.</p>

Presentation

Teaching Point	Behavioural Objectives	Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
2. Adaptation to Oceans	Student describes the adaptation to oceans.	<p>Teacher's Statement</p> <p>Marine animals like fish, dolphins etc. live in sea.</p> <p>* Most sea animals have streamlined bodies. fish, Dolphins have streamlined bodies.</p> <p>* Some sea animals such as squid and octopus do not have streamlined bodies. However, when they move in water, their bodies acquire streamlined shape.</p> <p>* They stay deeper in the sea than the seabed.</p>	Student listens carefully and writes in their notebook.	Lecture Method Demonstration Method.	Describe the adaptation to oceans
3. Adaptation to Ponds, lakes and Rivers	Student explains adaptation to Ponds, lakes and Rivers.	<p>Teacher's Statement</p> <p>Ponds & lakes are the habitats for small fish, frog and aquatic plants.</p> <p>* Aquatic plants of fresh water are called hydrophytes. Hydrophytes may be free floating, rooted & float or submerged. eg: lotus, water lily, Hydrilla</p> <p>* Some of the aquatic plants have their roots fixed in the soil under the water.</p>	Student listens carefully and notes down in their notebook.	Lecture Method.	

Class: VI Subject: Science
Topic: AQUATIC ADAPTATION IN ANIMALS & PLANTS

Aquatic plants of fresh water - HYDROPHYTES
eg: lotus, water lily, Hydrilla etc.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
		<p>* Some aquatic plants float on water.</p> <p>* The stems of aquatic plants are long, hollow & light.</p> <p>Hydrophytes adapt in the following ways:-</p> <p>i) In aquatic plants, roots are much reduced in size and their main function is to hold the plant in its place.</p> <p>ii) The stems may be reduced, long, slender, flexible and spongy. The air spaces in spongy stems make them buoyant.</p> <p>iii) Thin & large flat leaves with waxy surface, as in lotus & water lily. This helps them from wetting and rotting.</p>	<p>Student listens carefully and notes down in their notebook.</p>	<p>Boardwork</p> <p>Class: VI Subject: Science</p> <p>Topic: AQUATIC ADAPTATION IN ANIMALS AND PLANTS</p> <p><u>Hydrophytes</u></p> <ul style="list-style-type: none"> ↳ Roots - reduced in size. ↳ Stems - reduced/long/slender/flexible ↳ Leaves - thin 	<p>Explain the adaptation to Ponds, lakes and Rivers.</p>
4. Adaptation in frogs	Student describes the adaptation in frogs	<p>Teacher's Statement-</p> <p>Frogs usually have ponds as habitat. They can stay both in the pond water as well as on the land.</p> <p>Their webbed feet help them swim in water.</p> <p>Their strong back legs help in jumping and catching the prey.</p>	<p>Student listens carefully and notes down in their notebook.</p>	<p>Lecture Method.</p>	<p>Describe the adaptation in frogs.</p>

Closure

So students, today we discussed about aquatic adaptation in animals & plants

- adaptation to oceans
- adaptation to ponds, lakes & rivers
- adaptation in frogs.

Ques: Name 2 hydrophytes

Ques: Name one sea animal which does not have a streamlined body but when in water, its body acquires streamlined shape.

Home Assignment

Ques: Explain the adaptation in lotus plant.

Ques: How do dolphins and whales respire?

Assessment of Lesson

Date 5/2/2020
 of Intern: Julie Koshy
 Aquatic Adaptation in Animals & Plants Lesson Plan No. 20

Category Under Supervision	Rating Scale				
	Not Used	Poor	Avg.	Good	Very Good
Lesson Plan	0	1	2	3	4
Introduction	0	1	2	3	4
Presentation	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
Use of Teaching Aid	0	1	2	3	4
Chalk Board Work	0	1	2	3	4
Management of the Class	0	1	2	3	4
Evaluation	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4

all comments and suggestions:-

(Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Signature of the

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School of Education

Lesson Plan No. 21

Date 6.02.2020

Name of the Internship School National Victor Public School

Class & Section VI-B Period 3rd Duration 40 mins

Subject Science Sub-Subject Physics

Topic Electricity

Specific Objectives After my class, students will be able to

- cite the three types of power plants used for generating electricity.
- describe dry cell (battery cell) in detail.

Previous Knowledge

Students have general idea about the importance of electricity in day-to-day life.

Introduction

Ques: State few appliances used in everyday life which work with electricity?

Ans: Fans, Microwave, washing machine, television etc.

Ques: State any 2 dangers of electricity if handled carelessly.

Ans: Electric shock, can catch fire etc.

Ques: How is electricity generated?

Ans: No Response.

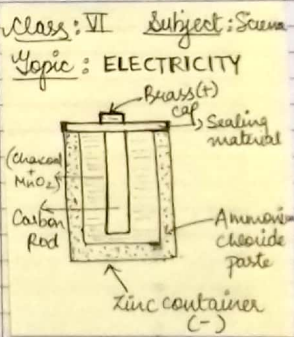
Announcement of the Topic

Today, we will discuss about 'Electricity'.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Electricity	Student cites the three types of power plants which are used for generating electricity on a large scale.	<p><u>Teacher's Statement</u></p> <p>Electricity plays an important role in our day-to-day life. It is a convenient and useful form of energy which is used to supply current for the use of heating, lighting etc.</p> <p>There are three types of power plants used for generating electricity on a large scale. These are:-</p> <ul style="list-style-type: none"> * Hydroelectric Power Plants (Hydel power plants) - These use kinetic energy of moving water. * Thermal Power Plants - Coal, Diesel or natural gas. * Atomic Power Plants - Nuclear fission of Uranium. <p>The electricity supplied to our homes by the electricity supply companies is the Alternating Current (AC). It comes from the power station.</p> <p>Ques: What is the source of electricity in torch?</p>	Students listen carefully and notes down in their notebook.	<p>Lecture Method</p> <p>Board work</p> <p>Class: VI Subject: Science Topic: ELECTRICITY</p> <p>3 Types of power plants for generating electricity:-</p> <ul style="list-style-type: none"> i) Hydroelectric Power Plants (use kinetic energy of moving water) ii) Thermal Power Plants (Coal, Diesel) iii) Atomic Power Plant (Nuclear fission of Uranium-235) 	
2. Electric Cell (Battery Cell)	Student describes dry cell in detail.	<p>To find out the source of electricity in a torch let's do the following activity.</p> <p>A torch is provided to the students and following instructions are being provided</p>	Students listen carefully and follow the instructions	<p>Lecture Method</p> <p>Question Answer Method</p>	

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
• Activity		<ul style="list-style-type: none"> - Open the screw cap. - What is there inside it? - Take these out. - Put the screw cap back on body of the torch. - Press the switch. - Does the bulb glow? - Put the cells inside the torch and replace the screw cap. - Now switch on the torch. - Does the bulb glow? <p>Yes, the bulb glows. The cells supply the required electricity to the bulb.</p>	<p>has battery cells inside it.</p> <p>the bulb doesn't glow.</p> <p>the bulb glows.</p>	<p>Question-Answer Method</p>	
• Description of a Dry Cell		<p><u>Teacher's Statement</u></p> <p>Dry cell (also called Battery or electric cell) is a source of electricity.</p> <p>It generates direct current due to chemical reaction that takes place inside it.</p> <p>It's available in different sizes.</p> <p>It was invented by G. Leclanché in 1868.</p> <p>- A dry cell consists of an outer container made of zinc. The zinc container acts as - terminal.</p> <p>A carbon rod having a brass cap located at the centre of the zinc container. This acts as its positive terminal.</p>	<p>student notes down in their notebook.</p>	<p>Lecture Method</p>	<p>Name the source of electricity in a torch.</p>
				<p>Class: VI Subject: Science Topic: ELECTRICITY</p> 	

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
		The carbon rod is surrounded by a mixture of manganese dioxide (MnO_2) and powdered charcoal.			
		The remaining space in the container is then filled with thick paste of ammonium chloride (NH_4Cl).			
		In a dry cell, the electric current flows from carbon electrode to zinc electrode.			Describe a dry cell in detail.
Advantages of a Dry cell.	Student cites the advantages of a Dry cell.	Teacher's Statement	Student listens carefully and notes down in notebook.	Lecture Method	
		Dry cells have the following advantages:-		<p>Class: VI Subject: Science</p> <p>Topic: ELECTRICITY</p> <p>Dry cells</p> <ul style="list-style-type: none"> ↳ light in weight ↳ small in size ↳ can be transported ↳ No leakage 	
		i) Dry cells are light in weight and small in size.			
		ii) Dry cells can be transported from one place to another easily.			
		iii) There is no fear of leakage in dry cells.			Cite the advantages of a Dry cell.

Closure

To students, today we discussed about
 - electricity generation
 - electric cell :- Description & Advantages

Ques: what is the direction of flow of current in a dry cell?

Ques: Name the source of electricity in a torch
 Draw a diagram showing the inner view of that source.

Home Assignment

Ques: Is human body a conductor or non-conductor of electricity?

Assessment of Lesson

of Intern: Julie Koshy Date: 6.02.2020
Electricity Lesson Plan No. 21

Category Under Supervision		Rating Scale				
		Not Used	Poor	Avg.	Good	Very Good
Lesson Plan Introduction	1. Objectives stated in behavioural terms	0	1	2	3	4
	2. Student's previous knowledge tested	0	1	2	3	4
	3. Topic was announced effectively	0	1	2	3	4
Presentation	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
	9. Skill of Illustration with examples	0	1	2	3	4
Use of Teaching Aid	10. Adequate & appropriate	0	1	2	3	4
	11. Quality of teaching aid	0	1	2	3	4
Chalk Board Work	12. Organized and adequate	0	1	2	3	4
	13. Appropriateness and clarity	0	1	2	3	4
Management of the Class	14. Class environment & discipline	0	1	2	3	4
	15. Supervision of student's work	0	1	2	3	4
Evaluation	16. Skill of Closure	0	1	2	3	4
	17. Based on the behavioral objectives	0	1	2	3	4
	18. Quality of home assignment	0	1	2	3	4

all comments and suggestions:-

(Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Signature of the

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School of Education

Lesson Plan No. 22

Date 7.02.2020

Name of the Internship School National Victor Public School

Class & Section VI-B Period 4th Duration 40 mins

Subject Science Sub-Subject Physics

Topic Lighting up a bulb with a dry cell.

Specific Objectives After my class, students will be able to

- describe the design of a torch bulb
- explain the right way to connect a bulb and a cell.

Previous Knowledge

Students have good knowledge about dry cell
- its structure and advantages.

Introduction

Ques: How does a bulb light up?

Ans: A bulb lights up when electricity is passed through it.

Ques: Name the source of electricity in a torch?

Ans: Dry cell.

Ques: How does the electricity pass from the cell to the bulb?

Ans: No response.

Announcement of the Topic

Today, we'll discuss about 'lighting up a bulb with a dry cell'.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Torch Bulb	Student describes the design of a torch bulb.	<p><u>Teacher's Statement</u></p> <p>Bulb is an electrical component used for lighting. Commonly used bulbs are based on the use of filament. The torch bulb works at low voltage (1.5 - 6 volts)</p> <p>The torch bulb consists of the following parts :-</p> <ol style="list-style-type: none"> Metal casing having grooves in the lower part. A small spiral of thin wire called the filament. Two slightly thicker wires. The upper ends of these wires support the filament. The lower end of one of the thin wires is connected to the metal casing and that of the other wire to the metal tip at the bottom of the bulb. The metal tip and the metal casing are fixed in such a way that they don't touch each other. This is done by using an insulation between the two 	Student listens carefully and notes down in their notebook.	<p>Lecture Method</p> <p>Boardwork</p> <p>Class: VI Subject: Science</p> <p>Topic: LIGHTING UP A BULB WITH A DRY CELL</p> <p>Torch bulb :-</p> <ul style="list-style-type: none"> ↳ Metal casing with grooves ↳ filament ↳ Insulation between metal tip & metal casing. 	Describe the design of a torch bulb.
2. Lighting up a bulb with a Dry cell	Student explain the right way to connect a bulb and a cell.	<p><u>Teacher's Statement</u></p> <p>A bulb will light up only when electricity is passed through it. A cell has 2 terminals and a bulb also has 2 terminals. If these terminals are connected rightly, the bulb should light up.</p>	Student listens carefully.	Lecture Method	

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
		<p>From the observations made in the activity described above the following conclusions can be drawn:-</p>	<p>Student notes down the observations in their notebook.</p>	<p>Boardwork Class: VI Subject: Science Topic: LIGHTING UP A BULB WITH A DRY CELL.</p> <p>Bulb lights up only when the cell, the bulb and the wires form a complete path.</p>	
		<p>* The bulb lights up only when the cell, the bulb and the wires form a complete path. This path starts from one terminal of the cell and ends at the other.</p>			
		<p>* The bulb doesn't light up if there is no gap in the path.</p>			
		<p>* The bulb also doesn't light up if the path starts and ends up at the same terminal of the cell.</p>			<p>Explain the right way to connect a bulb and a cell.</p>

Closure

So students, today we discussed about
 - the design of a torch bulb.
 - the right way to connect a bulb and a cell.

Ques:- Which part of the bulb glows when current is passed through it?

Ques:- Under which voltage does a torch bulb operate?

Home Assignment

Ques:- An electrician while handling any electrical appliance / circuit wears rubber shoes or rubber gloves. Why?

Assessment of Lesson

Name of Intern: Julie Koshy Date: 7-02-2020
 Topic: Lighting up a bulb with a dry cell Lesson Plan No.

Category Under Supervision	Rating Scale					
	Not Used	Poor	Avg.	Good	Very Good	
Lesson Plan Introduction	1. Objectives stated in behavioural terms	0	1	2	3	4
	2. Student's previous knowledge tested	0	1	2	3	4
	3. Topic was announced effectively	0	1	2	3	4
Presentation	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
	9. Skill of Illustration with examples	0	1	2	3	4
Use of Teaching Aid	10. Adequate & appropriate	0	1	2	3	4
	11. Quality of teaching aid	0	1	2	3	4
Chalk Board Work	12. Organized and adequate	0	1	2	3	4
	13. Appropriateness and clarity	0	1	2	3	4
Management of the Class	14. Class environment & discipline	0	1	2	3	4
	15. Supervision of student's work	0	1	2	3	4
Evaluation	16. Skill of Closure	0	1	2	3	4
	17. Based on the behavioral objectives	0	1	2	3	4
	18. Quality of home assignment	0	1	2	3	4

For all comments and suggestions:-
 (Classroom Management, Confidence, Voice modulation, Interaction with the students, Development of lesson with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Signature of the Mentor

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Lesson Plan No. 23

Date 10.02.2020

Name of the Internship School National Victor Public School

Class & Section VII-B Period 6th Duration 40 mins

Subject Science Sub-Subject Physics

Topic Electrical Circuits

Specific Objectives After my class, students will be able to

- define an electrical circuit.
- cite the difference between a closed & open circuit.
- state the components of an electrical circuit.

Previous Knowledge

Students have good knowledge about electrical appliances and their uses.

Introduction

Ques : How does a bulb light up?

Ans : A bulb lights up when electricity is passed through it.

Ques : Name the source of electricity in a torch.

Ans : Dry cell.

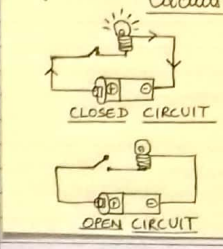
Ques : What is an electrical circuit?

Ans : No Response.

Announcement of the Topic

"Today, we'll discuss about 'Electrical Circuits'."

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Electrical Circuits	student defines an electrical circuit.	<p><u>Teacher's Statement</u></p> <p>The complete path of flow of electricity from one terminal of the cell to its other terminal through its various electrical components is called an electrical circuit.</p> <p>The circuit in which electric current flows from one terminal of a cell or battery to the other is called a closed circuit.</p> <p>In a closed circuit, current flows from the positive terminal of the cell to its negative terminal via various electrical components such as bulb etc.</p> <p>The circuit in which electrical contact at any point is broken is called an open circuit. No current flows in an open circuit.</p>	student listens carefully and notes down the important points in their notebook.	<p>Lecture Method</p> <p>Boardwork</p> <p>Class: VI Subject: Science Topic: electrical Circuits</p> 	
2. Fused Bulb does not glow.	Student explains the reason why a fused bulb doesn't glow.	<p><u>Teacher's Statement</u></p> <p>Bulb has a small filament inside it. As long as this filament is intact, it glows when connected to a cell.</p> <p>A bulb is said to be fused, when its filament is broken. The broken filament cannot complete the circuit.</p> <p>As a result, current cannot flow through the filament & the fused bulb doesn't glow.</p>	student listens carefully and writes in their notebook.	Lecture Method	explain why a fused bulb doesn't glow.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
3. Symbols of some electrical components.	Student depicts the components of an electrical circuit by its symbol.	<p>Teacher's Statement</p> <p>The components such as cells, bulbs, wires, switches etc. are called components of an electrical circuit.</p> <p>Each component of an electrical circuit can be described by its symbol.</p> <ul style="list-style-type: none"> Cell Battery Lamp Connecting Wire Switch \rightarrow Open \rightarrow Closed 	Student listens carefully and notes down the symbols in their notebook.	<p>Lecture Method</p> <p>Boardwork</p> <p>Class: VI Subject: Science Topic: Electrical Circuits</p> <p>Cell </p> <p>Battery </p> <p>Lamp </p> <p>Connecting wire </p> <p>Switch (open) </p> <p>Switch (closed) </p>	Depict the components of an electric circuit by its symbol.
4. Circuit Diagram	Student draws the circuit diagram of an electrical circuit.	<p>Teacher's Statement</p> <p>A diagram which shows the arrangement of various components in an electrical circuit with the help of their symbols is called a circuit diagram.</p>	Student draws the circuit diagram in their notebook.	<p>Lecture Method</p> <p>Class: VI Subject: Science Topic: Electrical Circuits</p> <p>Circuit Diagram showing an electrical circuit consisting of a bulb, a key and a dry cell.</p>	Draw the circuit diagram of an electric circuit.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
5. Electric Switch	Student defines an electric switch	<p><u>Teacher's Statement</u></p> <p>Switch is a device that makes or breaks a circuit. The on-off mechanism in an electrical circuit is controlled by a switch.</p> <p>* When the switch is in the 'off' position, the circuit is open and the current does not flow through the circuit.</p> <p>* When the switch is in the 'on' position, the circuit is closed and the current flows through the circuit.</p>	Student listens carefully and notes down in their notebook.	<p>Lecture Method.</p> <p>Class: VI Subject: Science Topic: Electrical Circuits</p> <p>SWITCH - A Device that makes or breaks a circuit 'off' position - circuit is open 'on' position - circuit is closed.</p>	Define an electric switch

Closure

So, students today we discussed about

- Electric Circuit
- Symbols of electrical components
- Circuit Diagram
- Electric Switch

Ques: Why does a fuse bulb not glow?

Ques: What is meant by a closed electrical circuit?

Home Assignment

Ques: How does electricity flow in an electric circuit? What is meant by an open circuit?

Assessment of Lesson

of Intern: Julie Koshy Date: 10-02-2020
Electrical Circuits Lesson Plan No. 23

Category Under Supervision	Rating Scale				
	Not Used	Poor	Avg.	Good	Very Good
Lesson Plan	0	1	2	3	4
Introduction	0	1	2	3	4
Presentation	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4
Use of Teaching Aid	0	1	2	3	4
Chalk Board Work	0	1	2	3	4
Management of the Class	0	1	2	3	4
Evaluation	0	1	2	3	4
	0	1	2	3	4
	0	1	2	3	4

all comments and suggestions:-
 s Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of n with the help of Students)

Julie

Signature of the Intern

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School of Education

Lesson Plan No. 24

Date 11-02-2020

Name of the Internship School National Victor Public School

Class & Section VIII-D Period 3rd Duration 40 mins

Subject Science Sub-Subject Chemistry

Topic Plastics

Specific Objectives After my class, students will be able to.

- state the characteristics of plastics.
- list the uses of plastics.
- classify plastics as thermoplastics & thermosetting plastics.

Previous Knowledge

Students have basic knowledge of plastics.

Introduction

Ques: What are these (shows a bottle, ^{pen, dustbin}) made of?

Ans: Plastic.

Ques: If I pour hot water into this bottle, what happens?

Ans: It gets deformed.

Ques: So plastics that can be deformed easily on heating are known as _____.

Ans: No Response.

Announcement of the Topic

"Today, we'll discuss about "Plastics".

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Plastics	Student explains about Plastics.	<p><u>Teacher's Statement</u></p> <p>It is a polymer formulated by hydrocarbons available in molecules like natural gas, oil, minerals and plants in which are either aligned in linear arrangement or in cross-link arrangement.</p>	Student listens carefully.	Lecture Method.	Explain about Plastics.
2. Characteristics of Plastics	Student states the characteristics of Plastics.	<p><u>Teacher's Statement</u></p> <p>Plastics possess the following unique characteristics :-</p> <ol style="list-style-type: none"> i) They have lighter weight and possess good strength. ii) They are inexpensive and find extensive use in domestic household industries. iii) They are non-reactive and are resistant to corrosion. iv) They are durable and hence can be moulded into any shape and size. v) They are poor conductor of heat and electricity. 	Student listens carefully and notes down in their notebook.	Lecture Method.	State the characteristics of Plastics.
3. Uses of Plastics	Student lists the uses of plastics.	<p><u>Teacher's Statement</u></p> <ol style="list-style-type: none"> i) They are extensively used for packaging tablets, threads, stitching, gloves & medical instruments. ii) Used to manufacture cookware due to their resistance to heat. iii) Teflon - A unique plastic used in making cookware. 	Student listens carefully.	Lecture Method.	List the uses of Plastics.

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
4. Classification of Plastics	Student classifies plastics into thermoplastic and thermosetting plastics.	<p><u>Teacher's Statement</u></p> <p>Plastics can be classified into</p> <ol style="list-style-type: none"> Thermoplastics and Thermosetting Plastics <p><u>Thermoplastics :-</u></p> <ul style="list-style-type: none"> - Deformed easily on exposure to heat and can also be bent easily. - Polyethylene, Polypropylene, polystyrene & PVC. - They find extensive use in making toys, electrical ins and can be polished well, hence often used in making vehicle windows and light covers. <p><u>Thermosetting Plastics :-</u></p> <ul style="list-style-type: none"> - Plastics which when moulded once, cannot be softened by heating. - Bakelite is a poor conductor of heat & electricity. It is used for making electric switches, handles of utensils. - Melamine is a versatile material. It resists fire & can tolerate heat better than other plastics. It is used for making floor tiles, kitchenware and furniture which resist fire. 	Student listens carefully and notes the two types of plastics in their notebook.	<p>Lecture, Method, Boardwork</p> <p>Class: VIII Subject: Science</p> <p>Topic: PLASTICS</p> <pre> graph TD A[PLASTICS] --> B[THERMOPLASTIC] A --> C[THERMOSETTING PLASTICS] B --> D[can be deformed easily & bent easily] C --> E[once moulded can't be softened] </pre>	
o Thermoplastics					
o Thermosetting Plastics					Classify plastics into two and differentiate between them.

Closure

To students, today we discussed about

→ Plastics - Characteristics, Uses and their classification.

Ques: Explain the difference between thermoplastic and thermosetting plastics.

Ques: Give examples to show that plastics are non-corrosive in nature.

Ques: Should the handle and brushes of a toothbrush be made of the same material? Explain.

Home Assignment

Ques: Explain why the following are made of thermosetting plastics:-

- i) Saucepan handles
- ii) Electric plugs / switches / plug boards

Assessment of Lesson

Name of Intern: Julie Koshy Date: 12/12/2020
 Topic: Plastics Lesson Plan No: 24

Category Under Supervision		Rating Scale				
		Not Used	Poor	Avg.	Good	Very Good
Lesson Plan	1. Objectives stated in behavioural terms	0	1	2	3	4
	2. Student's previous knowledge tested	0	1	2	3	4
Introduction	3. Topic was announced effectively	0	1	2	3	4
	4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
Presentation	5. Skill of Questioning	0	1	2	3	4
	6. Skill of Explanation	0	1	2	3	4
	7. Skill of Stimulus variation	0	1	2	3	4
	8. Skill of Reinforcement	0	1	2	3	4
	9. Skill of Illustration with examples	0	1	2	3	4
Use of Teaching Aid	10. Adequate & appropriate	0	1	2	3	4
	11. Quality of teaching aid	0	1	2	3	4
Chalk Board Work	12. Organized and adequate	0	1	2	3	4
	13. Appropriateness and clarity	0	1	2	3	4
Management of the Class	14. Class environment & discipline	0	1	2	3	4
	15. Supervision of student's work	0	1	2	3	4
Evaluation	16. Skill of Closure	0	1	2	3	4
	17. Based on the behavioural objectives	0	1	2	3	4
	18. Quality of home assignment	0	1	2	3	4

For all comments and suggestions:-
 (in Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of work with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Signature of the Mentor

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School of Education

Lesson Plan No. 25

Date 12/2/2020

Name of the Internship School National Victor Public School

Class & Section VII-A Period 2nd Duration 40 mins

Subject Science Sub-Subject Physics

Topic Reflection of Light

Specific Objectives After my class, students will be able to

- explain light
- describe incident ray, reflected ray & normal
- state the laws of reflection.

Previous Knowledge

Students have general idea about light.

Introduction

Ques : What is light?

Ans : Form of energy.

Ques : How does it travel?

Ans : Straight line.

Ques : The process of sending back the light rays which falls on the surface of an object is called _____.

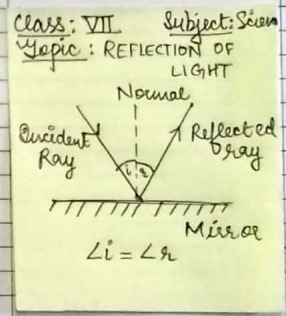
Ans : No Response.

Announcement of the Topic

Today, we'll discuss about "Reflection of Light".

Presentation

Teaching Point	Behavioural Objectives	Student Teacher's Activities	Students Expected Activities	Methods/Techniques, Teaching Aids & Chalk Board Work	Evaluation
1. Light	Student describes light.	<p>Teacher's Statement</p> <ul style="list-style-type: none"> - Radiation or form of energy that our eyes can detect. - light enables us to view our surroundings. - light travels from one place to another in a straight line. 	Student listens carefully.	Lecture Method.	Describe Light.
2. Incident Ray, Reflected Ray & Normal	Student cites the difference between incident ray & reflected ray.	<p>Teacher's Statement</p> <ul style="list-style-type: none"> o Incident Ray :- The light rays that fall on the reflecting surface. o Reflected Ray :- The light rays that get reflected back from a reflecting surface. o Normal :- It is a line that is perpendicular to the reflecting plane at the point of incidence of the incident ray. 	Student listens carefully and notes down in their notebook.	Lecture Method. Class: VII. Subject: Science Topic: REFLECTION OF LIGHT	Cite the difference between incident ray & reflected ray.
3. Laws of Reflection	Student states the laws of reflection.	<p>Teacher's Statement</p> <ul style="list-style-type: none"> i) The incident ray, the reflected ray & normal, all lie in the same plane. ii) Angle of Incidence = Angle of reflection $\angle i = \angle r$ 	Student listens carefully & notes down in their notebook.	Explanation Method.	State the laws of reflection.



Closure

To students, today we discussed about

- Reflection of Light
- Incident Ray, Reflected Ray & Normal
- Laws of Reflection.

Ques: State the laws of Reflection.

Ques: Why is angle of incidence always equal to angle of reflection?

Home Assignment

Ques: Find out the difference between concave & convex mirror with examples.

Assessment of Lesson

Name of Intern: Julie Koshy Date: 12/2/2020
 Topic: Reflection of light Lesson Plan No. 25

Category Under Supervision		Rating Scale					
		Not Used	Poor	Avg.	Good	Very Good	
A.	Lesson Plan	1. Objectives stated in behavioural terms	0	1	2	3	4
B.	Introduction	2. Student's previous knowledge tested	0	1	2	3	4
		3. Topic was announced effectively	0	1	2	3	4
		4. Systematic & logical development of the concept with the help of pupil	0	1	2	3	4
C.	Presentation	5. Skill of Questioning	0	1	2	3	4
		6. Skill of Explanation	0	1	2	3	4
		7. Skill of Stimulus variation	0	1	2	3	4
		8. Skill of Reinforcement	0	1	2	3	4
		9. Skill of Illustration with examples	0	1	2	3	4
D.	Use of Teaching Aid	10. Adequate & appropriate	0	1	2	3	4
		11. Quality of teaching aid	0	1	2	3	4
E.	Chalk Board Work	12. Organized and adequate	0	1	2	3	4
		13. Appropriateness and clarity	0	1	2	3	4
F.	Management of the Class	14. Class environment & discipline	0	1	2	3	4
		15. Supervision of student's work	0	1	2	3	4
G.	Evaluation	16. Skill of Closure	0	1	2	3	4
		17. Based on the behavioral objectives	0	1	2	3	4
		18. Quality of home assignment	0	1	2	3	4

Overall comments and suggestions:-

Dress Code, Chalk board entries, Confidence, Voice modulation, Interaction with the students, Development of lesson with the help of Students)

Julie

Signature of the Intern

Signature of the Mentor

Signature of the Mentor

Journal

(Description of self experience by the Intern after teaching)

Category Under Supervision			Rating Scale			
			Not Used	Poor	Avg.	Good
A.	Introduction	1. Student's previous knowledge tested	0	1	2	3
		2. Topic was announced effectively	0	1	2	3
B.	Presentation	3. Systematic & logical development of the concept with the help of pupil	0	1	2	3
		4. Skill of Questioning	0	1	2	3
		5. Skill of Explanation	0	1	2	3
		6. Skill of Stimulus variation	0	1	2	3
		7. Skill of Reinforcement	0	1	2	3
		8. Skill of Illustration with examples	0	1	2	3
C.	Use of Teaching Aid	9. Proper use of teaching aid(s)	0	1	2	3
		10. Tried to create interest in learning	0	1	2	3
D.	Chalk Board Work	11. Organized and adequate	0	1	2	3
		12. Appropriateness and clarity	0	1	2	3
E.	Management of the Class	13. Class environment & discipline	0	1	2	3
		14. Supervision of student's work	0	1	2	3
F.	Evaluation	15. Skill of Closure	0	1	2	3
		16. Based on the behavioral objectives	0	1	2	3
		17. Quality of home assignment	0	1	2	3

Over all experience and suggestion for my future teaching from myself

(Eg.- I satisfied the queries of the students. Today I feel satisfied/unsatisfied with my teaching.)

Julie

Signature of the Intern

Signature of the N