School of Medical and Allied Sciences

Pharmacy ETE - Jun 2023

Time: 3 Hours **Marks**: 75

Sem IV - BP403T / BPHT4003

Physical Pharmaceutics II Theory
Your answer should be specific to the question
asked Draw neat labeled diagrams wherever necessary

1.	Recall types of colloids.	K1 CO3 (2)
2.	Explain the term thixotropy.	K2 CO3 (2)
3.	Interpret about the term Lyophillic and Lyophobic colloids.	K2 CO1 (2)
4.	List out kinetic properties of colloid particles.	K1 CO4 (2)
5.	Interpret about the Newtonian system with suitable example.	K2 CO2 (2)
6.	What is Zero order reaction explain with formula?	K1 CO5 (2)
7.	What are Colloid and Colloidal dispersion?	K1 CO1 (2)
8.	Summarize the term adsorption of solids.	K2 CO4 (2)
9.	Define dilatant flow of liquids.	K1 CO2 (2)
10.	Extend the phenomenon of Isoelectric point.	K2 CO5 (2)
11)	Build a note on applications of colloids.	K3 CO1 (5)
OR		
	Build a note on Tyndall effect.	K3 CO1 (5)
12.	Identify the term surface active agent and HLB scale with examples.	K3 CO3 (5)
13.	Analyze about the kinetic properties of colloid (Sedimentation and Osmotic pressure).	K4 CO1 (5)
14.	Simplify the term Thixotropy and write in detail about thixotropy in dilatant system. OR	K4 CO2 (5)
	Analyze Plastic flow of material with rheogram.	
15.	Plan a note on micro and macroemulsions with example.	K3 CO2 (5)
16)	Contrast on types of suspension and emulsions.	K4 CO3 (5)
OR		
	Compare flocculated and deflocculated suspension.	K4 CO3 (5)
17.	Examine the term NDDS and classify different NDDS techniques.	K6 CO6 (5)
18.	Justify the title Communition (Size reduction) and explain in detail about Colloidal mill.	K5 CO4 (10)
19)	Develop a note on Accelerated stability studies related to pharmaceutical products.	K6 CO5 (10)
OR		
	Discuss about stability studies and factors which influence the stability characters of pharmaceutical products.	K6 CO5 (10)